ACS CHEMISTRY FINAL EXAM STUDY GUIDE

- milli (m) (m has 3 lines, upside down it's a w (win) so negative=loss): 10^-3
- micro (inapropppiate joke, i don't want Quizlet to ban me lol): 10^-6
- nano: 10^-9
- pico (p): 10^-12
- mono: one, single, alone
- di: two
- tri: three (tricycle)
- tetra: four (tetrapod) dogs, cats, vertebrae animals
- penta: five (pentagon)
- hexa: 6 (hexagon)
- hepta: 7 (heptagon)
- octa: 8 (octopus, stop)
- nona: 9 (no = neign in German)
- deca: 10 (decade = 10 years)
- molar mass: the mass of one mole of a pure substance
- molar ratio: ratio of coefficients in a chemical equation
- limiting reactant: A reactant that is totally consumed during a chemical reac-tion, limits the extent of the reaction, and determines the amount of product.
- percent composition: the percent by mass of each element in a compound
- molarity: the number of moles of solute per liter of solution
- percent yield: actual yield/theoretical yield x 100
- empirical formula: a chemical formula showing the ratio of elements in acompound rather than the total number of atoms
- molecular formula: A chemical formula that shows the number and kinds of atoms in a
 molecule, but not the arrangement of the atoms.
- gas laws: describe how the behavior of gases is affected by pressure, volume, and temperature

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Puja Mehta

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Learning Science Through Computer Games and Simulations National Research Council, Division of Behavioral and Social Sciences and Education, Board on Science Education, Committee on Science Learning: Computer Games,

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Study Guide for Chemistry by Steven S. Zumdahl Martha B. Barrett,1986 Crimes Against the English Language, Study Guide to accompany Basic Concepts of Chemistry, 7th Edition Leo J. Malone,2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry Study Guide for Chang's Chemistry Kenneth W. Watkins,1988 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Students Study Guide Fredeen, Pearson, 2003-07

Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry Katherine J. Denniston,1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in

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Complete Preparation for the AHPAT, 1999 Williams & Wilkins Review, 1998-04 Here is the most accurate and complete resource designed for students interested in applying for admission to health fields such as medical technology and medical therapy Reviews cover science knowledge verbal and quantitative ability and reading comprehension Students will value the topic outlines for the AHPAT exam the in depth allied health glossary and the full length sample test complete with solutions

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