

## Chapter 10 test

**Do I want to be a manager?**

Match each item with the correct statement below.

- |                            |                                      |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition               |
| b. mole                    | e. standard temperature and pressure |
| c. Avogadro's number       | f. empirical formula                 |

1. the number of representative particles of a substance present in 1 mole of that substance
2. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
3. the SI unit used to measure amount of substance
4.  $0^{\circ}\text{C}$  and 1 atm
5. the percent by mass of each element in a compound
6. the smallest whole number ratio of the atoms in a compound

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- [illegible]

# Chapter 10 Chemistry Test

**Linda D. Williams**



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**Industrial and Engineering Chemistry**, 1925 **Industrial & Engineering Chemistry**, 1923 **The Phlebotomy Textbook** Susan King Strasinger, Marjorie Schaub Di Lorenzo, 2019-02-25 Rely on this comprehensive resource to master the techniques you need to safely obtain quality specimens You'll understand all the hows and whys that lead to success in this rapidly changing field Inside you'll find the up to date coverage of routine procedures and their complications as well specialized procedures quality and infection control state of the art equipment medical terminology ethical and legal issues body systems and related diagnostic laboratory tests

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