

CHEMISTRY - CHAPTER 3: EQUATIONS



BALANCING CHEMICAL EQUATIONS - NAMES GIVEN Sheet #2

Practice

1. Potassium reacts with water yielding potassium hydroxide and hydrogen
2. Chlorine reacts with potassium bromide yielding potassium chloride and bromine
3. Zinc + hydrochloric acid yields zinc chloride and hydrogen
4. iron + water $\rightarrow \text{Fe}_3\text{O}_4$ + hydrogen
5. zinc sulfide + oxygen \rightarrow zinc oxide + sulfur dioxide
6. $\text{C}_2\text{H}_6 + \text{Cl}_2 \rightarrow \text{C} + \text{HCl}$
7. Aluminum + sodium hydroxide $\rightarrow \text{NaAlO}_2$ + hydrogen
8. $\text{C}_2\text{H}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
9. Ammonia + oxygen \rightarrow nitrogen monoxide + water
10. Phosphorus + iron(III) oxide \rightarrow tetraphosphorus decoxide + iron
11. Copper sulfide + oxygen \rightarrow copper(II) oxide + sulfur dioxide
12. sodium bicarbonate + hydrogen sulfate \rightarrow sodium sulfate + water + carbon dioxide
13. Sodium carbonate + silver nitrate \rightarrow sodium nitrate + silver carbonate
14. Calcium + oxygen \rightarrow calcium oxide
15. Zinc + ferric oxide \rightarrow zinc oxide + iron

Chapter 11 Chemical Reactions Worksheet

G Thomas



Chapter 11 Chemical Reactions Worksheet:

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Understanding Chemistry Fred M. Dewey,1994

Chemistry Stanley R. Radel,Marjorie H. Navidi,1990

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Structure and Change Alan C.

Blizzard,Richardson, R. A. (Robert A.),Humphreys, D. A. (David A.),1981

Study Guide for Chemical Principles [by] Steven

S. Zumdahl Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

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