Physical Science- Chapter 24 Test Study Guide KEY

1	 the substances that are about to react.
	Reactants
2	- the new substances produced.
	Products
2 T =	aw of Conservation of Mass- the French alchemist Antoine Lavoisier established that the mass of the
	lucts always the mass of the reactants.
per son	Equals
4	a way to describe a chemical reaction using chemical formulas and
othe	er symbols.
	Chemical Equation
5.74) means?
	Solid
6. (a	q) means?
	Aqueous, dissolved in water
-	means?
	Yields, produces, forms
	areas, produces, forms
9. Tì	Created, destroyed he numbers to the left of formulas for reactants and products are called
	Coefficients
10.	- represents the number of units of each substance taking part in a reaction.
	Coefficients
LEG	The following underlined numbers are called NiCl ₂ (aq) + 2NaOH (aq) - Ni(OH) ₂ (s) + 2NaCl (aq)
	Coefficients
	Coefficients
12.1	fron exposed to and corrodes or rusts forming iron (III)
pocied	e, rust can seriously damage iron structures because it crumbles and exposes more iron to the air.
	Air, moisture
13.0	Copper corrodes when exposed to the air forming a blue-green coating called a Patina
14.	- has the same number of atoms of each element on both sides of
	equation.
	Balanced chemical equation

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Kara Sealock, Cydnee Seneviratne

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