Names		

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- 13. If the spin of one electron in an orbital is clockwise, what is the spin of the other electron in that orbital?
  - a. more

c. counterclockwise

b. clockwise

- both clockwise and counterclockwise
- 14. What types of atomic orbitals are in the third principal energy level?
  - a. s and p only.

c. s, p, and d only

b. p and d only

- d. x.p.d. and f
- 15. What is the next atomic orbital in the series 1s, 2s, 2p, 3s, 3p?
  - a. 2db. 3d

- c. 3*j*
- 16. What is the number of electrons in the outermost energy level of an oxygen atom?
  - a. 3

b. 4

- 6
- 17. What is the electron configuration of potassium?
  - a. 1s<sup>2</sup>2s<sup>2</sup>2p<sup>2</sup>3s<sup>2</sup>3p<sup>2</sup>4s<sup>1</sup>

c. 1s22s23s23p63d6

b. 1s<sup>2</sup>2s<sup>2</sup>2p<sup>10</sup>3s<sup>2</sup>3p<sup>3</sup>

- d. 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>1</sup>
- 18. How many unpaired electrons are in a sulfur atom (atomic number 16)?
  - a. 0

b. 1

c. 2

- d. 3
- 19. Which is the correct electron dot diagram for the element Silicon (Si)?









a. b.

- e.
- 20. How many valence electrons are present in silicon?
  - a. Two

b. Four

Six

d. Eight

Completion: Complete each statement.

- The \_\_\_\_\_\_ principle states that a maximum of two electrons may occupy a single atomic orbital, but only if the electrons have opposite spins.
- 22. The arrangement of electrons in an atom is called the \_\_\_\_\_\_ of an atom,
- The element \_\_\_\_\_ has the following electron configuration: [Rn]5/6a<sup>n</sup>7x<sup>2</sup>.
- 24. The f-block elements are also referred to as the lanthanides and \_\_\_\_\_\_.
- 25. Xenon has \_\_\_\_\_\_\_ valence electrons.
- 26. Aluminum has \_\_\_\_\_\_\_valence electrons.

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Joan I. Senyk, Darrell D. Ebbing

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