

WKS
Naming & Formulas of Covalent
Compounds & Acids WS

NAME _____
Period _____ **Date** _____

1. What is a binary molecular compound?

A covalently-bonded compound composed of only two kinds of non-metal elements, that does not start with H (those compounds are acids).

2. In the table below, fill in the prefix corresponding to the number of atoms in the formula:

Number	Prefix	Number	Prefix
1	mono-	6	hexa-
2	di-	7	hepta-
3	tri-	8	octa-
4	tetra-	9	nona-
5	penta-	10	deca-

In the table below, determine the formula of the binary covalent compound or acid from its name, or the name of the compound from the formula.

Name to Formula		Formula to Name	
3. antimony tribromide	SbBr_3	14. P_4S_5	tetraphosphorous pentasulfide
4. iodine pentafluoride	IF_5	15. SeO_3	selenium trioxide
5. dinitrogen trioxide	N_2O_3	16. Si_2Br_6	disilicon hexabromide
6. ammonia	NH_3	17. SCL_4	sulfur tetrachloride
7. phosphorus triiodide	PI_3	18. CH_4	methane
8. carbon monoxide	CO	19. NF_3	nitrogen trifluoride
9. phosphorous pentachloride	PCl_5	20. N_2O	dinitrogen monoxide
10. tetraiodine nonoxide	I_4O_9	21. IF_7	iodine heptafluoride
11. bromic acid No hydro, so bromate	HBrO_3	22. HClO	ClO^- = hypochlorite hypochlorous acid
12. phosphorous acid No hydro, so phosphite	H_3PO_3	23. HF	F^- = fluoride hydrofluoric acid
13. hydroiodic acid hydro, so iodide	HI	24. H_2CO_3	CO_3^{2-} = carbonate carbonic acid

25. What is the difference between a binary acid and an oxyacid? How are each named?

A binary acid contains H covalently bonded to what looks like a monatomic anion from the periodic table (or a polyatomic anion ending in -ide). An oxyacid contains H covalently bonded to what looks like a polyatomic oxyanion from Table D. The binary acid is named by adding "hydro-" before the root and "-ic acid" after the root of the anion, and oxyacids are named by changing the ending of the anion, either "-ate" to "-ic acid" or "-ite" to "-ous acid"

Ch 8 2 Acids Covalent Naming Answer Key

Kathleen P. Talaro, Arthur Talaro



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Organic Chemistry Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts *Organic Chemistry An Acid Base Approach* provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules

Brown's Introduction to Organic Chemistry William H. Brown, Thomas Poon, 2017-06-28 *Introduction to Organic Chemistry* 6th Global Edition provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major It is most suited for a one semester organic chemistry course In an attempt to highlight the relevance of the material to students the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science particularly the biological and health sciences The text illustrates the use of organic chemistry as a tool in these sciences it also stresses the organic compounds both natural and synthetic that surround us in everyday life in pharmaceuticals plastics fibers agrochemicals surface coatings toiletry preparations and cosmetics food additives adhesives and elastomers

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Table of Contents Ch 8 2 Acids Covalent Naming Answer Key

1. Understanding the eBook Ch 8 2 Acids Covalent Naming Answer Key
 - The Rise of Digital Reading Ch 8 2 Acids Covalent Naming Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ch 8 2 Acids Covalent Naming Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ch 8 2 Acids Covalent Naming Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ch 8 2 Acids Covalent Naming Answer Key
 - Personalized Recommendations
 - Ch 8 2 Acids Covalent Naming Answer Key User Reviews and Ratings

- Ch 8 2 Acids Covalent Naming Answer Key and Bestseller Lists
- 5. Accessing Ch 8 2 Acids Covalent Naming Answer Key Free and Paid eBooks
 - Ch 8 2 Acids Covalent Naming Answer Key Public Domain eBooks
 - Ch 8 2 Acids Covalent Naming Answer Key eBook Subscription Services
 - Ch 8 2 Acids Covalent Naming Answer Key Budget-Friendly Options
- 6. Navigating Ch 8 2 Acids Covalent Naming Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ch 8 2 Acids Covalent Naming Answer Key Compatibility with Devices
 - Ch 8 2 Acids Covalent Naming Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ch 8 2 Acids Covalent Naming Answer Key
 - Highlighting and Note-Taking Ch 8 2 Acids Covalent Naming Answer Key
 - Interactive Elements Ch 8 2 Acids Covalent Naming Answer Key
- 8. Staying Engaged with Ch 8 2 Acids Covalent Naming Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ch 8 2 Acids Covalent Naming Answer Key
- 9. Balancing eBooks and Physical Books Ch 8 2 Acids Covalent Naming Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ch 8 2 Acids Covalent Naming Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ch 8 2 Acids Covalent Naming Answer Key
 - Setting Reading Goals Ch 8 2 Acids Covalent Naming Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ch 8 2 Acids Covalent Naming Answer Key
 - Fact-Checking eBook Content of Ch 8 2 Acids Covalent Naming Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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