

CHAPTER 11, Chemical Reactions *(continued)*

9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction. Hydrochloric acid reacts with zinc to produce aqueous zinc(II) chloride and hydrogen gas.

**► Balancing Chemical Equations** (pages 324–328)

10. What is the law of conservation of mass?

In any physical or chemical change, mass is neither created nor destroyed.

11. Complete the flowchart for balancing equations.

Determine the correct formulas and physical states for the _____ **reactants** _____ and _____ **products** _____.



Write a _____ **skeleton equation** _____ with the formulas for the reactants on the left and the formulas for the products on the right of a yields sign (—→).



Count the number of _____ **atoms** _____ of each element in the reactants and in the products.



Balance the number of atoms of the elements on the two sides of the equation by placing _____ **coefficients** _____ in front of formulas. Never try to balance an equation by changing the _____ **subscripts** _____ in formulas.



Check each atom or polyatomic ion to be sure the equation is _____ **balanced** _____, and make sure that all coefficients are in the _____ **lowest** _____ possible ratio.

Chapter 11 1 Chemical Reactions Workbook Answers

Society of Chemical Industry (Great Britain)



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Organic Synthesis Michael Smith, 2024-08-13 Organic Synthesis 5e provides a reaction based approach to this important branch of organic chemistry Updated and accessible this eagerly awaited revision offers a comprehensive foundation for graduate students coming from disparate backgrounds and knowledge levels to provide them with critical working knowledge of basic reactions stereochemistry and conformational principles This reliable resource uniquely incorporates molecular modeling content problems and visualizations and includes reaction examples and homework problems drawn from the latest in the current literature There have been advancements in organic reactions particularly organometallic reactions and there is a need to show how these advancements have influenced current organic synthesis The goal is to revise and update the examples of reaction examples taken from the synthesis literature from about 2017 2023 The reactions illustrate those that are used most often in modern organic synthesis but recent examples will show their current relevance Where new approaches and new reactions have been developed for organic synthesis examples will be added as new material Provides new content reaction examples and study problems from recent research Features improved organization new art and new chapter content on process chemistry and green organic chemistry Includes revised homework for each chapter with new examples and questions

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