

Physics

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(Chapter 11) (Thermodynamics) (Exercises)

(Class 11)

Question 11.1:

A geyser heats water flowing at the rate of 3.0 litres per minute from 27 °C to 77 °C. If the geyser operates on a gas burner, what is the rate of consumption of the fuel if its heat of combustion is 4.0×10^4 J/g?

Answer 11.1:

Water is flowing at a rate of 3.0 litre/min.

The geyser heats the water, raising the temperature from 27°C to 77°C.

Initial temperature, $T_1 = 27^\circ\text{C}$ and Final temperature, $T_2 = 77^\circ\text{C}$

Rise in temperature, $\Delta T = T_2 - T_1 = 77 - 27 = 50^\circ\text{C}$

Heat of combustion = 4×10^4 J/g

Specific heat of water, $c = 4.2 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$

Mass of flowing water, $m = 3.0 \text{ litre/min} = 3000 \text{ g/min}$

Total heat used, $\Delta Q = mc \Delta T = 3000 \times 4.2 \times 50 = 6.3 \times 10^5 \text{ J/min}$

\therefore Rate of consumption = $\frac{6.3 \times 10^5}{4 \times 10^4} = 15.75 \text{ g/min}$

Question 11.2:

What amount of heat must be supplied to 2.0×10^{-2} kg of nitrogen (at room temperature) to raise its temperature by 45 °C at constant pressure? (Molecular mass of $N_2 = 28$; $R = 8.3 \text{ J mol}^{-1} \text{ K}^{-1}$.)

Answer 11.2:

Mass of nitrogen, $m = 2.0 \times 10^{-2} \text{ kg} = 20 \text{ g}$ and Rise in temperature, $\Delta T = 45^\circ\text{C}$

Molecular mass of N_2 , $M = 28$

Universal gas constant, $R = 8.3 \text{ J mol}^{-1} \text{ K}^{-1}$

Number of moles, $n = \frac{m}{M} = \frac{2.0 \times 10^{-2} \times 10^3}{28} = 0.714$

Molar specific heat at constant pressure for nitrogen $C_p = \frac{7}{2}R = \frac{7}{2} \times 8.3 = 29.05 \text{ J mol}^{-1} \text{ K}^{-1}$

The total amount of heat to be supplied is given by the relation: $\Delta Q = nC_p \Delta T$

$= 0.714 \times 29.05 \times 45 = 933.38 \text{ J}$

Therefore, the amount of heat to be supplied is 933.38 J.

Question 11.3:

Explain why

- (a) Two bodies at different temperatures T_1 and T_2 if brought in thermal contact do not necessarily settle to the mean temperature $(T_1 + T_2)/2$.
- (b) The coolant in a chemical or a nuclear plant (i.e., the liquid used to prevent the different parts of a plant from getting too hot) should have high specific heat.
- (c) Air pressure in a car tyre increases during driving.
- (d) The climate of a harbour town is more temperate than that of a town in a desert at the same latitude.

Answer 11.3:

- (a) When two bodies at different temperatures T_1 and T_2 are brought in thermal contact, heat flows from the body at the higher temperature to the body at the lower temperature till equilibrium is achieved, i.e., the temperatures of both the bodies become equal. The equilibrium temperature is equal to the mean temperature $(T_1 + T_2)/2$ only when the thermal capacities of both the bodies are equal.

Chapter 11 Assessment Physics

Katrin Zwirgmaier



Chapter 11 Assessment Physics:

Psychology, Pedagogy, and Assessment in Serious Games Connolly, Thomas M., Hainey, Thomas, Boyle, Elizabeth, Baxter, Gavin, Moreno-Ger, Pablo, 2013-11-30 This book addresses issues the potential of games to support learning and change behaviour offering empirical evidence pertaining to the effectiveness of Serious Games in the key areas of psychology pedagogy and assessment

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Solar Terrestrial Physics: Present and Future, 1984 Solar Terrestrial Physics D. M. Butler, Konstantinos Papadopoulos, 1984

Phy P&P Les Plans Blk Sch 99 Zitzewitz, 1998-06 **Earth Science**, 2001 **Focus on Earth Science**, 2001

Computed Tomography - E-Book Euclid Seeram, 2015-09-02 Build the foundation necessary for the practice of CT scanning with Computed Tomography Physical Principles Clinical Applications and Quality Control 4th Edition Written to meet the varied requirements of radiography students and practitioners this two color text provides comprehensive coverage of the physical principles of CT and its clinical applications Its clear straightforward approach is designed to improve your understanding of sectional anatomic images as they relate to CT and facilitate communication between CT technologists and other medical personnel Comprehensively covers CT at just the right depth for technologists going beyond superficial treatment to accommodate all the major advances in CT One complete CT resource covers what you need to know The latest information on advances in CT imaging including advances in volume CT scanning CT fluoroscopy multi slice applications like 3 D imaging CT angiography and virtual reality imaging endoscopy all with excellent coverage of state of the art principles instrumentation clinical applications and quality control More than 600 photos and line drawings help students understand and visualize concepts Chapter outlines show you what is most important in every chapter Strong ancillary package on Evolve facilitates instructor preparation and provides a full complement of support for teaching and learning with the text NEW Highlights recent technical developments in CT such as the iterative reconstruction detector updates x ray tube innovations radiation dose optimization hardware and software developments and the introduction of a new scanner from Toshiba NEW Learning Objectives and Key Terms at the beginning of every chapter and a Glossary at the end of the book help you organize and focus on key information NEW End of Chapter Questions provide opportunity for review and greater challenge NEW An added second color aids in helping you read and retain pertinent information Ate Science Plus 2002 LV Red Holt Rinehart & Winston, 2001-02 Technical Abstract Bulletin, 1978 **Instructing and Testing Advanced Learners** Slava Kalyuga, 2006 When we learn new and complex materials our cognitive processing capabilities are usually

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Radar in Meteorology David Atlas, 2015-03-30. This fully illustrated volume covers the history of radar meteorology, deals with the issues in the field from both the operational and the scientific viewpoint, and looks ahead to future issues and how they will affect the current atmosphere. With over 200 contributors, the volume is a product of the entire community and represents an unprecedented compendium of knowledge in the field.

Glencoe Algebra 1, 2001. *Algebra 1* Foster, McGraw-Hill/Glencoe, 1996-04-08. A beginning algebra textbook.

Assessment in Higher Education John Heywood, 1977. Heywood teacher education, Trinity College Dublin, builds on his previous contributions to the assessment debate by examining assessment not only of student learning but also of institutions, the programs they offer, and the teaching they provide. He details the significant developments that have taken place over the last decade in the field and clarifies the many different meanings of the term assessment that are now in use. He primarily addresses those involved in designing and implementing assessment but also considers issues that arise for policy maker and managers. Distributed in the US by Taylor and Francis.

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Algebra 1 Collins Publishers Staff, William Collins, Gilbert J. Cuevas, Russell J. Foster, Gordon Staff, Moore-Harris, Rath, William L. Swart, Winters, 1998.

Science Advice to NASA Joseph K. Alexander, 2017. This book documents highlights of NASA's interactions with outside scientific advisors over the agency's full lifetime and draws lessons from that history for research managers, decision makers, and scientists. The book is divided into three parts: the first two being focused on history and the third on synthesis and analysis. Part 1 briefly examines early forerunner activities at NACA and in the decade leading up to NASA's formation, and it then considers NASA's use of outside advice during its first three decades. Part 2 picks up the story in 1988 and follows it up to 2016. Part 3 examines a sampling of case studies, discusses recurring characteristics of notably successful advisory activities, and provides a glimpse at what past experience might imply for the future of scientific advice at NASA. The last two chapters provide big picture summaries of themes that have emerged from earlier discussions.

New York Math A, 2004.

The E-Assessment Handbook Geoffrey Crisp, 2007-09-05. Assessing learning in an online environment is being used by teachers and institutions at an increasing rate.

This text provides practical guidance on various aspects of online assessment including types of assessment choosing the right software and examples of e assessments over a wide variety of disciplines Sigma Xi Quarterly Society of the Sigma Xi, 1929

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