MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

 Formation of solutions where the process is endothermic can be spontaneous prov 	vided that	- 60
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- A) the solvent is a gas and the solute is a solid
- B) they are accompanied by an increase in order
- C) they are accompanied by an increase in disorder
- D) they are accompanied by another process that is exothermic
- E) the solvent is water and the solute is a gas-
- The plurase "like dissolves like" refers to the fact that _______
 - A) polar solvents dissolve nonpolar solutes and vice versa.
 - B) solvents can only dissolve solutes of similar molar mass
 - C) gases can only dissolve other gases
 - D) polar solvents dissolve polar solutes and nonpolar solvents dissolve nonpolar solutes
 - E) condensed phases can only dissolve other condensed phases
- 3) An unsaturated solution is one that _______,
 - A) has no double bonds.
 - B) has a concentration lower than the solubility
 - C) contains no solute.
 - D) contains the maximum concentration of solute possible, and is in equilibrium with undissolved solute
 - E) contains more dissolved solute than the solubility allows.
- A solution with a concentration higher than the solubility is _______
 - A) is not possible.
 - B) is saturated.
 - C) is supercritical.
 - D) is unsaturated
 - E) is supersaturated
- 5) Which one of the following substances would be the most soluble in CCl4?
 - A) H₂O

B) NaCL

- C) CH3CH2OH
- D) NH₃.

E) C10H22

- 6) Which one of the following is most soluble in water?
 - A) CH-OH
 - B) CH₃CH₂CH₂OH
 - C) CH₂CH₂OH
 - D) CH₃CH₂CH₂CH₂CH₂OH
 - E) CH₃CH₂CH₂CH₂OH

Chapter 11 Chemistry Test Answers

Robert R. Redfield

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