#### **Justin Silva**

1) Indicate the oxidation number of carbon and sulfur in the following compounds.

a. 
$$CO = (C = +2)$$

b. 
$$CO2 = (C = +4)$$

c. 
$$Na2CO3 = (C = +4)$$

d. 
$$Na2C2O4 = (C = +3)$$

e. 
$$CH4 = (C = -4)$$

f. 
$$H2CO = (C = 0)$$

g. 
$$SO2 = (S = +4)$$

$$b. SO3 = (S = +6)$$

i. 
$$Na2SO4 = (S = + 6)$$

$$j. Na2SO3 = (S = +4)$$

$$k. Na2S2O3 = (S = +2)$$

1. 
$$Na2S4O6 = (S = +5/2 = +2.5)$$

$$m. SC12 = (S = +2)$$

$$n. Na2S2 = (S = -1)$$

o. 
$$SOC12 = (S = +4)$$

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**Albert A Gayle** 

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