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$$C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$$
l construct a table showing how to interpret the information in this equation in terms of a single molecule of glucose moles of reactants and products

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web there's 2 on the left and 1 on the right so we need to change the coefficient of NH_3 to 2 now we have $N_2 + H_2 \rightarrow 2NH_3$ total the atoms up again on the left there is 2 N and 2 H still on the right there is 2 N and 6 H now so now all we need to do is make the left side have 6 H in total

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