

# ***Carbon & Its Compounds***

**Class- VIII**

**By: M.A.Shamsi**

**Fill in the blanks by choosing the correct option:**

1. An element with atomic number 6 and has got ability to form large number of compounds \_\_\_\_\_ (boron/ carbon)
2. Hydrocarbons are the compounds made up Carbon & \_\_\_\_\_ (nitrogen /hydrogen)
3. Oxides of carbon are \_\_\_\_\_ compounds. (organic / inorganic)
4. In free state carbon occurs in the form of a shiny material called \_\_\_\_\_ (diamond / graphite)
5. In  $\text{CaCO}_3$  carbon occurs in \_\_\_\_\_ state. ( free/ combined)
6. The ability of an element to exist in more than one physical form is termed as \_\_\_\_\_ (molecularity/ allotropy)
7. Allotropes have \_\_\_\_\_ physical properties but \_\_\_\_\_ chemical properties. (Similar /different).
8. Fullerene is \_\_\_\_\_ form of carbon. (crystalline/ amorphous)
9. Coal is \_\_\_\_\_ form of carbon. (crystalline/ amorphous)
10. Buckminster fuller ( $\text{C}_{60}$ ) has \_\_\_\_\_ carbon atoms in its sphere. (30/ 60/ 90)

# Carbon Compounds Answer Key

**Society of Chemical Industry (Great Britain)**



## **Carbon Compounds Answer Key:**

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web 3 is the note dated signed and legible 4 is the client name and or identifier included on each page 5 has referral and collateral information been documented 6 does the note reflect changes in client status eg gaf measures of functioning 7 are all abbreviations standardized and consistent 8 did counselor supervisor sign note 9

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