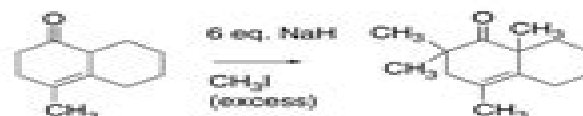


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH_2 in liquid NH_3 . The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH_2 in liquid NH_3 .



1. The nitrile anion was formed using NaNH_2 in liquid NH_3 . The nitrile anion was then alkylated with PhCH_2Cl to form the nitrile alkylated product.
2. The nitrile alkylated product was then hydrolyzed to form the nitrile acid. The nitrile acid was then decarboxylated to form the nitrile. The nitrile was then reduced to form the nitrile amine. The nitrile amine was then alkylated with PhCH_2Cl to form the nitrile alkylated product.



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