CH 317 Chromatography Practice Questions with answers

 The following data were obtained by gas-liquid chromatography on a 40 cm packed column:

Compound	ts: (min)	W (at base) (min)
Unretained Compound	1.9	-
Methylcyclohexane	10.0	0.76
Methylcyclohexene	10.9	0.82
Toluene	13.4	1.06

Calculate

(a) An average number of plates from the data.

$$N_A = 16(10.0/0.76)^2 = 2770$$

$$N_B = 16(10.9/0.82)^2 = 2827$$

$$N_C = 16(13.4/1.06)^2 = 2557$$

$$N_{AVG} = (2770 + 2827 + 2557) / 3 = 2718 = 2.7 \times 10^3 \text{ plates}$$

(b) An average plate height for the column.

$$L = NH$$

$$H = 40 \text{ cm} / 2.7 \times 10^3 \text{ plates} = 0.0148 \text{ cm/plate} = 0.015 \text{ cm/plate}$$

(c) The resolution for methylcyclohexene and methylcyclohexane.

$$R = 2(10.9 - 10.0) / (0.82 + 0.76) = 1.139 = 1.14$$

(d) The resolution for methylcyclohexene and toluene.

$$R = 2(13.4 - 10.9) / (1.06 + 0.82) = 2.66$$

(e) The resolution for methylcyclohexane and toluene.

$$R = 2(13.4 - 10.0) / (1.06 + 0.76) = 3.74$$

Chromatography Questions And Answers

M Lipman

Chromatography Questions And Answers:

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Table of Contents Chromatography Questions And Answers

- 1. Understanding the eBook Chromatography Questions And Answers
 - The Rise of Digital Reading Chromatography Questions And Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chromatography Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chromatography Questions And Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chromatography Questions And Answers
 - Personalized Recommendations
 - Chromatography Questions And Answers User Reviews and Ratings
 - Chromatography Questions And Answers and Bestseller Lists

- 5. Accessing Chromatography Questions And Answers Free and Paid eBooks
 - o Chromatography Questions And Answers Public Domain eBooks
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 - Chromatography Questions And Answers Compatibility with Devices
 - Chromatography Questions And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Chromatography Questions And Answers
 - Interactive Elements Chromatography Questions And Answers
- 8. Staying Engaged with Chromatography Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chromatography Questions And Answers
- 9. Balancing eBooks and Physical Books Chromatography Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chromatography Questions And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chromatography Questions And Answers
 - Setting Reading Goals Chromatography Questions And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chromatography Questions And Answers
 - Fact-Checking eBook Content of Chromatography Questions And Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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