Hints to Balancing Chemical Equations

- Adjust the coefficients not the subscripts.
- Even/Odd Multiply by 2 to make all even.
- Balance polyatomic ions as a whole.
- Check the balanced equation.

Example:

$$ZnS + O_2 \Rightarrow ZnO + SO_2$$
 (O is even on the left and odd on the right)

$$ZnS + O_2 \Rightarrow 2ZnO + SO_2$$
 (multiply by 2 to make O even)

Example:

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Basic Chemistry William S. Seese, Guido H. Daub, 1977

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