Name: \_\_\_\_\_ Date: \_\_\_\_

#### CHEMISTRY BALANCING CHEMICAL REACTIONS

Type the correct coefficient in front of each reactant and product to make the balanced equation. Only type a number. No spaces before or after.

$$1 \quad \underline{\hspace{1cm}} Na_{(s)} + \underline{\hspace{1cm}} Cl_{2(g)} \rightarrow \underline{\hspace{1cm}} NaCl_{(s)}$$

2 
$$Na_{(s)} + O_{2(g)} \rightarrow Na_2O_{(s)}$$

3 
$$Mg(s) + O_{2(g)} \rightarrow MgO(s)$$

4 \_\_\_\_Mg(s) + \_\_\_\_HCl<sub>(aq)</sub> 
$$\rightarrow$$
 \_\_\_\_MgCl<sub>2(aq)</sub> + \_\_\_\_H<sub>2(g)</sub>

5 
$$H_2O_{2(aq)} \rightarrow H_{2(g)} + O_{2(g)}$$

6 
$$Al_{(s)} + HF_{(aq)} \rightarrow AlF_{3(aq)} + H_{2(g)}$$

7 
$$NaClO_{3(aq)} \rightarrow NaCl_{(aq)} + O_{2(g)}$$

8 
$$ZnS_{(s)} + ZnS_{(s)} + Zn_3P_2(s) + Zn_3P_2(s)$$

9 
$$Ag_2S_{(s)} \rightarrow Ag_{(s)} + S_{8(s)}$$

10 \_\_\_\_ Ba(NO<sub>3</sub>)<sub>2(aq)</sub> + \_\_\_\_ H<sub>2</sub>CO<sub>3(aq)</sub> 
$$\rightarrow$$
 \_\_\_\_ BaCO<sub>3(s)</sub> + \_\_\_\_ HNO<sub>3(aq)</sub>

11 \_\_\_\_ Pb(OH)<sub>4(aq)</sub> + \_\_\_\_ Cu<sub>2</sub>O<sub>(aq)</sub> 
$$\rightarrow$$
 \_\_\_\_ PbO<sub>2(s)</sub> + \_\_\_ CuOH<sub>(aq)</sub>

12 \_\_\_\_ 
$$H_3PO_{4(aq)} +$$
 \_\_\_\_  $HCl_{(aq)} \rightarrow$  \_\_\_\_  $PCl_{5(s)} +$  \_\_\_\_  $H_2O_{(l)}$ 

13 
$$C_2H_{6(s)} + O_{2(g)} \rightarrow CO_{2(g)} + H_2O_{(g)}$$

14 \_\_\_\_ 
$$C_3H_{8(s)}$$
 + \_\_\_\_  $O_{2(g)}$   $\rightarrow$  \_\_\_\_  $CO_{2(g)}$  + \_\_\_\_  $H_2O_{(g)}$ 

15 
$$C_5H_{10(s)} + O_{2(g)} \rightarrow CO_{2(g)} + H_2O_{(g)}$$

# **Balance And Identify Chemical Reactions**

**David M. Hanson** 

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