Balancing Chemical Equations

Balance the equations below.

3)
$$Z_{\text{NaCl}} + \underline{\qquad} F_2 \rightarrow \underline{\qquad} NaF + \underline{\qquad} Cl_2$$

4)
$$2 H_2 + 0_2 \rightarrow 2 H_2 O$$

5) ____ Pb(OH)₂ +
$$\frac{2}{2}$$
 HCl \rightarrow $\frac{2}{2}$ H₂O + ____ PbCl₂

6)
$$2 \text{ AlBr}_3 + 7 \text{ K}_2 \text{SO}_4 \rightarrow 6 \text{ KBr} + Al_2 (\text{SO}_4)_3$$

7) ___ CH₄ +
$$\frac{2}{3}$$
 O₂ \rightarrow ___ CO₂ + $\frac{2}{3}$ H₂O

8)
$$C_0H_6 + 5_{02} \rightarrow 3_{C02} + 4_{H_20}$$

9)
$$C_0H_{10} + 2.5O_2 \rightarrow 6.0O_2 + 9.0O_2$$

13)
$$2 Ag_2O \rightarrow 4 Ag + 0_2$$

16)
$$\frac{2}{K} + \frac{2}{MgBr} \rightarrow \frac{2}{KBr} + \frac{2}{MgBr}$$

17)
$$2 \text{HCI} + 2 \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + 2 \text{CO}_2$$

19)
$$Z_{H_2O} + Z_{O_2} \rightarrow Z_{H_2O_2}$$

20)
$$\mathbb{Z}$$
NaBr + \mathbb{Z} CaF₂ $\rightarrow \mathbb{Z}$ NaF + \mathbb{Z} CaBr₂

21) ____ H₂SO₄ +
$$\sum$$
 NaNO₂ \rightarrow \sum HNO₂ + ___ Na₂SO₄

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Virginia P. Powell

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