Name: _____

Date: _____

Balancing Chemical Equations

Balance the following chemical equations.

1.
$$\underline{2}$$
 Fe + $\underline{3}$ H₂SO₄ \longrightarrow $\underline{1}$ Fe₂(SO₄)₃ + $\underline{3}$ H₂

2.
$$\underline{1}$$
 CH₄ + $\underline{2}$ O₂ \longrightarrow $\underline{1}$ CO₂ + $\underline{2}$ H₂O

3.
$$1$$
 SiCl₄(ℓ) + 2 H₂O(ℓ) \longrightarrow 1 SiO₂(s) + 4 HCl(aq)

4.
$$\underline{2}$$
 AgI + $\underline{1}$ Na₂S \longrightarrow $\underline{1}$ Ag₂S + $\underline{2}$ NaI

5.
$$\underline{4}$$
 NH₃ + $\underline{5}$ O₂ \longrightarrow $\underline{4}$ NO + $\underline{6}$ H₂O

6. 1 FeO₃(s) + 3 CO(g)
$$\longrightarrow$$
 1 Fe(ℓ) + 3 CO₂(g)

7.
$$\underline{1}$$
 SiO₂ + $\underline{4}$ HF \longrightarrow $\underline{1}$ SiF₄ + $\underline{2}$ H₂O

9.
$$4 (NH_4)_3PO_4 + 3 Pb(NO_3)_4 \longrightarrow 1 Pb_3(PO_4)_4 + 12 NH_4NO_3$$

Balancing Chemical Equations Unit 4 Answer Key

John T. Moore, Richard H. Langley

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