Balancing Chemical Reactions

Balance the following equations:

4.
$$N_2 + O_2 \rightarrow N_2O_5$$

5.
$$_H_3PO_4 + _Mg(OH)_2 \Rightarrow _Mg_3(PO_4)_2 + _H_2O$$

7. ___H₂ + ___O₂
$$\rightarrow$$
 ___H₂O₂

10.
$$Cs + N_2 \rightarrow Cs_2N$$

11.
$$\underline{\hspace{1cm}}$$
 Mg + $\underline{\hspace{1cm}}$ Cl₂ \Rightarrow $\underline{\hspace{1cm}}$ MgCl₃

12. ___Rb + ___RbNO₈
$$\Rightarrow$$
 ___Rb₂O + ___N₂

13. ___C_H_0 + ___O_
$$\rightarrow$$
 ___CO_ \rightarrow + ___H_0O

15. __
$$C + __ O_2 \rightarrow __ CO_2$$

16.
$$C_2H_0 + C_2 - C_2 + H_0O$$

17.
$$\underline{\hspace{1cm}}$$
 ii + $\underline{\hspace{1cm}}$ AlCl₃ \Rightarrow $\underline{\hspace{1cm}}$ LiCl + $\underline{\hspace{1cm}}$ Al

$$10. \underline{\hspace{1cm}} Rb + \underline{\hspace{1cm}} P \Rightarrow \underline{\hspace{1cm}} Rb_S P$$

19.
$$Al(OH)_1 + H_1SO_1 \Rightarrow Al_2(SO_1)_1 + H_2O_1$$

20. Rb +
$$S_6 \rightarrow Rb$$
, S

21.
$$CaSO_4 + KOH \rightarrow Ca(OH)_2 + K_2SO_4$$

23.
$$Al_4C_8 + _H_2O \Rightarrow _CH_4 + _Al(OH)_8$$

24. __Ba + __IINO₃
$$\Rightarrow$$
 __II₂ + __Ba(NO₃)₃

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Charles Loudon Bloxam

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