

## Identifying Nuclear Reaction Equations

### Key?

Students must identify the product(s) of a nuclear reaction. Using currently balanced equations to represent various nuclear reactions is not the best method. All equations need to be balanced to maintain correct conservation laws. It is more difficult to construct, and the distribution library is restricted.

### Reaction Criteria

- Chemical conservation laws are strictly followed as in chemical reactions/equations.
- Atomic number and mass number are used to construct a balanced equation.

### Pre-equations

- Various symbols for identifying nuclear reactions, particle numbers, and isotopes.

### Equations

- $\alpha$  is helium nucleus, electron is  $e^-$  and alpha particles.

- $\beta^+$  means a positron, and  $\beta^-$  means an electron (a  $\beta^-$  particle is the same as an electron).

### Notes

The following are balanced nuclear equations:



### Key Questions

- What are the products of sodium-23 decay? What are the masses of the products?  
The alpha particle ( ${}_{2}^4\text{He}$ ) and  $\text{Na-23}$  ( ${}_{11}^{23}\text{Na}$ ).
- What are the products of sodium-23 decay? What are the masses of the products?  
A beta particle ( ${}_{-1}^0\text{e}^-$ ) and  $\text{Ne-23}$  ( ${}_{10}^{23}\text{Ne}$ ).

# Balancing Nuclear Reaction Equations Pogil

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**Process Oriented Guided Inquiry Learning (POGIL)** Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL's theoretical basis its implementations in diverse environments and evaluation of student outcomes [Science Citation Index](#), 1992 Vols for 1964 have guides and journal lists [Tables of Nuclear Reaction Q Values](#) Val J. Ashby, Henry C. Catron, 1959 *Nuclear Reactions*, 1957

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