

- _____ 1. A chemical reaction is a process in which
 a. products change into reactants c. substances change state
 b. the law of conservation of mass applies d. all of these
- _____ 2. During a chemical reaction,
 a. new elements are produced c. atoms are destroyed
 b. atoms are rearranged d. elements are destroyed
- _____ 3. An equation is balanced by
 a. changing subscripts c. erasing elements as necessary
 b. adding coefficients d. adding elements as necessary
- _____ 4. An atom's ability to undergo chemical reactions is determined by its
 a. protons b. innermost electrons c. neutrons d. outermost electrons
- _____ 5. What are the reactants in the following chemical equation:

$$\text{Zn} + \text{CuSO}_4 \longrightarrow \text{ZnSO}_4 + \text{Cu}$$
 a. zinc and copper c. zinc and copper (II) sulfate
 b. zinc sulfate and copper d. only zinc
- _____ 6. What are the products in the above equation?
 a. zinc and copper c. zinc and copper (II) sulfate
 b. zinc sulfate and copper d. only zinc

For questions 7-12, classify the reaction according to the type it is. Put that answer in the blank. Then add coefficients to balance the reaction when necessary.



Balancing Chemical Equations Multiple Choice Answer

S Marginson



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