

## Balancing Nuclear Reaction Equations

1989-90

Students must have been taught an efficient way. Using systematic techniques requires no special or other special knowledge. Nuclear reactions are balanced by balancing the number of protons and neutrons on each side of the equation. The number of protons is the atomic number, and the number of neutrons is the mass number.

### Example 1: Balancing

- a. The following reaction has been written by someone who is not a chemist. Balance it.
- b. Write a balanced nuclear equation for the following reaction.

### Example 2: Balancing

- a. The following reaction has been written by someone who is not a chemist. Balance it.

### Example 3: Balancing

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1989-90 The following reaction has been written by someone who is not a chemist. Balance it.

### Example 4: Balancing

The following reaction has been written by someone who is not a chemist. Balance it.



### Example 5: Balancing

- a. The following reaction has been written by someone who is not a chemist. Balance it.

- b. The following reaction has been written by someone who is not a chemist. Balance it.

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Example 1: The following reaction has been written by someone who is not a chemist. Balance it.

# Balancing Nuclear Reactions Answers Pogil

**R Barnett**



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