Balance the following equations by adding coefficients before the elements or compounds in the following equations.

1. 
$$\_NH_4NO_3 \rightarrow \_N_2O + \_H_2O$$

3. 
$$\_O_2 \rightarrow \_O_3$$

4. 
$$\_Mg + \_HCl \rightarrow \_MgCl_2 + \_H_2$$

5. 
$$\_SO_3 \rightarrow \_SO_2 + \_O_2$$

6. 
$$\underline{Cu} + \underline{S_8} \rightarrow \underline{Cu_2S}$$

7. 
$$Na_2O + H_2O \rightarrow NaOH$$

8. 
$$PCl_5 + H_2O \rightarrow H_3PO_4 + HCl$$

9. 
$$\_FeO + \_O_2 \rightarrow \_Fe_2O_3$$

10. 
$$NaOH + H_2SO_4 \rightarrow Na_2SO_4 + H_2O$$

11. 
$$\_Al + \_Cl_2 \rightarrow \_AlCl_3$$

12. 
$$\_NH_3 + \_Na \rightarrow \_H_2 + \_NaNH_2$$

13. 
$$\_C_9H_{20} + \_O_2 \rightarrow \_CO_2 + \_H_2O$$

14. 
$$\_CaSiO_3 + \_HF \rightarrow \_SiF_4 + \_CaF_2 + \_H_2O$$

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Rajan Keshri

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Physical Chemistry in Depth Johannes Karl Fink, 2009-09-16 Physical Chemistry in Depth is not a stand alone text but complements the text of any standard textbook on Physical Chemistry into depth having in mind to provide profound understanding of some of the topics presented in these textbooks Standard textbooks in Physical Chemistry start with thermodynamics deal with kinetics structure of matter etc The Physical Chemistry in Depth follows this adjustment but adds chapters that are treated traditionally in ordinary textbooks inadequately e g general scaling laws the graphlike structure of matter and cross connections between the individual disciplines of Physical Chemistry Admittedly the text is loaded with some mathematics which is a prerequisite to thoroughly understand the topics presented here However the mathematics needed is explained at a really low level so that no additional mathematical textbook is needed

Chemistry in Quantitative Language Christopher O. Oriakhi, 2021 Chemistry in Quantitative Language is an invaluable guide to solving chemical equations and calculations It provides readers with intuitive and systematic strategies to carry out the many kinds of calculations they will meet in general chemistry This book provides innovative intuitive and systematic strategies to tackle any type of calculations encountered in chemistry Each chapter introduces the basic theories and concepts of a particular topic focusing on relevant equations Worked examples illuminate each type of problem with carefully explained step by step solutions Since chemistry problem can be presented in a number of ways the examples include several versions of each questions To help students understand and retain the procedures the solutions discuss not only what steps to carry out to reach solutions but why The second edition contains additional problems at the end of each chapter with varying degrees of difficulty and many of the original examples have been revised Book jacket Computational, Education, and Materials Science Aspects Ponnadurai Ramasami, 2022-10-03 Chapters collected from The Virtual Conference on Chemistry and its Applications VCCA 2021 Research and Innovations in Chemical Sciences Paving the Way Forward This conference was held in August 2021 and organized by the Computational Chemistry Group of the University of Mauritius These peer reviewed chapters offer insights into research on fundamental and applied chemistry with interdisciplinary subject matter **Problem Solving in General Chemistry** Ronald A. DeLorenzo,1981 The Merck Report ,1894

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