Name: Hour: Date:

#### Chemistry: Balancing & Classifying Chemical Equations

<u>Directions</u>: First, balance each of the chemical equations below. Then, classify each reaction as **synthesis**, **decomposition**, **single-replacement**, or **double-replacement**. To earn full credit, write the words out when classifying.

#### Balance the equation...

...and classify it.

10. \_\_\_ Pb(NO<sub>3</sub>)<sub>2</sub> + \_\_\_ H<sub>2</sub>S 
$$\rightarrow$$
 \_\_\_ PbS + \_\_\_ HNO<sub>3</sub>

19. \_\_\_ Al + \_\_\_ 
$$H_2SO_4 \rightarrow$$
 \_\_\_  $Al_2(SO_4)_3 + ___ H_2$ 

# **Balancing Chemical Equations Webquest Answers**

**Alan Trevor Neuff** 

### **Balancing Chemical Equations Webquest Answers:**

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