

balancing Equations

Name: _____

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Page _____

Introduction/Questions

- Write the equation for the reaction of sodium and chlorine (chlorine gas) to form sodium chloride.



- Write the equation for the reaction of sodium nitrate and sodium chloride to produce sodium chloride and sodium nitrate.



Information: Subscripts and Coefficients

It is important to understand that atoms are never really destroyed or created. For example, in Ca_3P_2 there are 3 calcium atoms and 2 phosphorus atoms. In the product, CaCl_2 , there are 3 calcium atoms and 2 phosphorus atoms.

A coefficient tells how many molecules of a substance are involved in a reaction. For example, in the equation $2\text{H}_2\text{O}$, the 2 is the coefficient. The 2 tells us that there are 2 molecules of water present. In the equation $2\text{H}_2\text{O}$, there are a total of 4 hydrogen atoms and 2 oxygen atoms.

Balancing Chemical Equations

- Balance the equation above the reaction in Ca_3P_2 . 3/1
- Balance the equation above the reaction in Ca_3P_2 .



Information: How To Balance Equations

Consider the reaction of sodium and chlorine to produce sodium chloride. This equation is:



Remember that when chlorine is in Cl_2 it is always written as Cl_2 . On the product side of the equation, there are 2 atoms of chlorine. So, on the product side, there is only one atom of sodium. To make sure the equation is balanced, we need to balance the equation. This can be done by adding a "2" to the product side:



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Rajan Keshri



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Balancing Chemical Equations Workbook Kraaya Publishing, 2021-02-21 Chemical Reactions to Balance Workbook This chemistry balancing equations practice workbook contains 250 non balanced chemical equations Begin with 2 terms problems Work your way up to 6 terms problems This is the perfect workbook to increase chemistry balancing skills for beginners Table of contents How To Balance A Chemical Equation Chemical Equations To Balance Correct Answers Book features Non repetitive equations Include all reactions types synthesis combustion decomposition Use it now and develop instant recall of balancing equations Enjoy the challenge

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