

Ionic and Covalent Naming Practice **Answers**

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

- | | | |
|-----|---------------------------|----------------------------------|
| 1) | Na_2CO_3 | sodium carbonate |
| 2) | P_4O_{10} | tetraphosphorus decaoxide |
| 3) | N_2O_5 | dinitrogen pentoxide |
| 4) | FeSO_4 | iron (II) sulfate |
| 5) | SiO_2 | silicon dioxide |
| 6) | FeCl_3 | iron (III) chloride |
| 7) | CoBr_2 | cobalt (II) bromide |
| 8) | B_2H_4 | diboron tetrahydride |
| 9) | CO | carbon monoxide |
| 10) | PCl_3 | phosphorus trichloride |

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

- | | | |
|-----|------------------------|--|
| 11) | dinitrogen trioxide | N_2O_3 |
| 12) | nitrogen dioxide | NO_2 |
| 13) | sulfur pentoxide | SO_5 |
| 14) | lithium acetate | $\text{LiC}_2\text{H}_3\text{O}_2$ |
| 15) | phosphorus trifluoride | PF_3 |
| 16) | vanadium (V) oxide | V_2O_5 |
| 17) | aluminum hydroxide | $\text{Al}(\text{OH})_3$ |
| 18) | zinc sulfide | ZnS |
| 19) | silicon tetrafluoride | SiF_4 |
| 20) | silver phosphate | Ag_3PO_4 |

Covalent Compounds Test Answer Key

**Antony C. Wilbraham, Dennis D.
Staley, Michael S. Matta, Edward L.
Waterman, Prentice-Hall Staff**

Covalent Compounds Test Answer Key:

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Table of Contents Covalent Compounds Test Answer Key

1. Understanding the eBook Covalent Compounds Test Answer Key
 - The Rise of Digital Reading Covalent Compounds Test Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Compounds Test Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Compounds Test Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Compounds Test Answer Key
 - Personalized Recommendations
 - Covalent Compounds Test Answer Key User Reviews and Ratings
 - Covalent Compounds Test Answer Key and Bestseller Lists

5. Accessing Covalent Compounds Test Answer Key Free and Paid eBooks
 - Covalent Compounds Test Answer Key Public Domain eBooks
 - Covalent Compounds Test Answer Key eBook Subscription Services
 - Covalent Compounds Test Answer Key Budget-Friendly Options
6. Navigating Covalent Compounds Test Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Covalent Compounds Test Answer Key Compatibility with Devices
 - Covalent Compounds Test Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Compounds Test Answer Key
 - Highlighting and Note-Taking Covalent Compounds Test Answer Key
 - Interactive Elements Covalent Compounds Test Answer Key
8. Staying Engaged with Covalent Compounds Test Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Compounds Test Answer Key
9. Balancing eBooks and Physical Books Covalent Compounds Test Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Compounds Test Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Covalent Compounds Test Answer Key
 - Setting Reading Goals Covalent Compounds Test Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Covalent Compounds Test Answer Key
 - Fact-Checking eBook Content of Covalent Compounds Test Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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