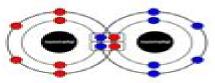
# DRAWING COVALENT BONDS

#### Introduction:

A covalent bond forms between nonmetals. Covalent bonds involve sharing pairs of electrons between two atoms so that each atom gains a full outer valence shell. Sharing 1 pair of electrons creates a single covalent bond, sharing 2 pairs of electrons creates a double covalent bond, and sharing 3 pairs of electrons creates a triple covalent bond.



	How to Draw Simple Covalent Bond Diagrams						
1.	Draw the electron dot diagrams for each element.  HINT: This will help you see the number of unpaired valence electrons for each element.						
2.	Count the total number of valence electrons between all atoms.  HINT: The total number of valence electrons should be 8 in most cases, except 2 hydrogens. If there are not 8, add more atoms of the element with the least number of valence electrons to start.						
3.	Determine which element is the central atom in the molecule, if needed.  HINT: The central atom is typically the atom with the largest valence number or the atom that there is only one of in the formula. If there are only 2 atoms in the formula, there is no central atom?						
4.	Pair up single electrons from different atoms to form covalent bonds and circle each pair.  HIMT: Keep doing this until each single electron from all atoms have been paired up. You may need to use double or triple bonds to make this work. This step takes trial and error. If one arrangement is not working, try adding another element or trying a new structure. Practice makes perfect!						

#### Activity: Drawing Covalent Bond Diagrams:

- For each covalent compound, you have been given the element names and the chemical formula.
- Follow the steps listed above to create a covalent bonding diagram for the compound.
- Remember, this takes trial and error. The first 2 have been done for you as examples.

#	Nonmetal	Nonmetal	Formula	Bonding Diagram
	hydrogen	chlorine	HCI	
A.	H • (must share 1)	CI: (must share 1)	(Formula shows 1 H and 1 Cl)	H ©CI:
В.	hydrogen	oxygen	H <sub>2</sub> O	
	H • (must share 1)	(must share 2)	(Formula shows 2 H and 1 O)	н⊜⊜∶

# **Covalent Bonding Workbook Answers**

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## Enjoying the Song of Appearance: An Mental Symphony within Covalent Bonding Workbook Answers

In a world used by screens and the ceaseless chatter of instantaneous communication, the melodic beauty and mental symphony created by the written term usually diminish into the background, eclipsed by the relentless noise and distractions that permeate our lives. However, located within the pages of **Covalent Bonding Workbook Answers** an enchanting literary treasure filled with natural emotions, lies an immersive symphony waiting to be embraced. Crafted by an elegant composer of language, this fascinating masterpiece conducts visitors on a psychological journey, well unraveling the concealed tunes and profound impact resonating within each cautiously constructed phrase. Within the depths of the emotional analysis, we shall discover the book is main harmonies, analyze their enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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