

Total

ame:	
arte:	

Total CaClO<sub>1</sub>

Total Al(OH)<sub>a</sub> ement.

Total (NH<sub>4</sub>),PO<sub>4</sub>

Total Al<sub>2</sub>(SO<sub>3</sub>),

Total

# of Atoms

## emota onithu WORKSHEET

- Chemical formulas can have three components.
  - a) The \_\_\_\_\_\_ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
  - which are numbers that are found on the lower of each element symbol. This tells you the number of \_\_\_\_\_\_ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only atom of this element in the molecule.
  - c) The which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be \_\_\_\_\_\_ by the \_\_\_\_\_ on the outside of - hand side of that bracket. the bracket found on the lower \_\_\_
- 2. atoms in the molecule:

Total

1) NaOH		2) HNO,		3) Li <sub>2</sub> O
Element	# of Atoms	Element	# of Atoms	Element
				- 2
Total		Total		Total
4) Li <sub>2</sub> SO <sub>4</sub>		5) NH <sub>4</sub> CI		6) CaClO
Element	# of Atoms	Element	# of Atoms	Element
Total		Total		Tot
7) H <sub>2</sub> COCH <sub>2</sub>		8) Mg(OH)		9) Al(OH
Element	# of Atoms	Element	# of Atoms	Element
Total		Total		Total
10) NH <sub>4</sub> C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>		11) NaC <sub>2</sub> H <sub>3</sub> C	D <sub>2</sub>	12) (NH <sub>4</sub> ),
Element	# of Atoms	Element	# of Atoms	Element
Total		Total		Total
13) Ca <sub>1</sub> (PO <sub>4</sub> ) <sub>2</sub>		14) Fe <sub>2</sub> (CO <sub>3</sub> )	h	15) Al <sub>2</sub> (SO
Element	# of Atoms	Element	# of Atoms	Element

# **Counting Atoms Grade 9**

Thirumalaisamy P. Velavancorresponding

### **Counting Atoms Grade 9:**

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session of approximately 60 post deadline papers We wish to thank all of the participants for their outstanding contributions and for preparation of their papers now available to a wider audience Our thanks go to the members of the International Steering Committee for their suggestions and recommendations We are especially pleased to have held this conference under the auspices of the International Union of Pure and Applied Physics VICOLS would not have been possible with out the financial support of the Natural Sciences and Engineering Research Council of Canada and the Office of Naval Research and Air Force Office of Scientific Research of the United States of America Scientific American, 1878 Energy ENC Focus, 2002 Three Cognitive Skills in Research Abstracts ,1984 The New International Year Book ,1921 New International Yearbook ,1921 Chemistry and Their Application to Stoichiometry Ardas Ozsogomonyan, 1977 The Readers' Guide to Periodical Literature ,1927 Nature Sir Norman Lockver, 1924 Summary ,1906 Internal Revenue Record and Customs Journal ,1892 The Library Index to Periodicals and Current Events ,1905 Catalogue of Printed Books British Museum, 1891

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