

# SCH 2202 ORGANIC CHEMISTRY II

► JANUARY-APRIL 2024

► JKUAT- DEPT OF CHEMISTRY

## BENZENE AND AROMATICITY

- **Benzene** ( $C_6H_6$ ) is the simplest **aromatic hydrocarbon** (or arene).
- Four degrees of unsaturation.
- It is planar.
- All C—C bond lengths are equal.
- Whereas unsaturated hydrocarbons such as alkenes, alkynes and dienes readily undergo addition reactions, benzene does not.



- Benzene reacts with bromine only in the presence of  $FeBr_3$  (a Lewis acid), and the reaction is a substitution, not an addition.



## STRUCTURE OF BENZENE

- The true structure of benzene is a resonance hybrid of the two Lewis structures.

The hybrid

Some texts draw benzene as a hexagon with an inner circle:

# Course Outline Of Sch 2202 Organic Chemistry Two Ku

**Timothy R. Rettich, David N.  
Bailey, Jeffrey A. Frick, Forrest J. Frank**

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

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

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