Name: Date:

## Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? They can share the markers.

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They can give the ones they need to each other.

#### Gizmo Warm-up

Just like the students described above, nonmetal atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play ( b) to see the electrons orbiting the nucleus of each atom.



The outermost electrons in each atom are called valence electrons. How many valence

Click Pause (III). Drag an electron from the left atom to the right atom. Click Play.

What happens? \_\_They share an electron.

Click Pause, drag an electron from the right atom to the left, and then click Play.

What happens now? They share another one.

# **Covalent Bonds Gizmo Quiz Answers**

**Victor M. Corman** 

**Covalent Bonds Gizmo Quiz Answers:** 

#### Covalent Bonds Gizmo Quiz Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Covalent Bonds Gizmo Quiz Answers**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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