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Quiz & Worksheet - Double Displacement Reaction

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1. What type of reaction is the following?

$$PbCl_2(aq) + K_2SO_4(aq) \longrightarrow 2KCl(aq) + PbSO_4(s)$$

e combustion

er gas formation

precipitation

neutralization

2. What are the ions and the correct charges for the following compound:

PbCl₂

Pb2+ Cl2-

- Pb2- Cl2-
- " Pb+ Cl2-

" Pb2+ Cl-

 What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)

Na₂SO₄+ 2AgNO₃ →

Ag₂SO₄ + Na₂NO₃

" Ag₂SO₄+ NaNO₃

Ag₂SO₄+ 2NaNO₃

AgSO₄ + NaNO₃

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Double Replacement Reactions Lab 2 Answer Key

David Kirk

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