ATUNGRUCTURE

a guided reading

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Amornio Maron

Ellery Smooth

Administration Studentiers

Afonde Structure

There are over 100 different types of atoms, called elements. Each seem is made up of these smaller particles called protocos, neutrons, and electrons.



Peopless and sewiness are found in the center of the atom in a sporte called the nucleus. Proteins are positively charged, and neutrons do not have a charge-they are neutral. Section of this, the nucleus has an overall positive charge. Ernest Rutherford discovered this in 1899.

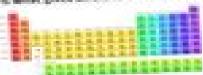
when he observed that other positively charged particles were repetited by the nucleus, because proteins and neutrons are much heavier than electrons, these particles make up the alsoesic mass of the atom.

Electronis are found outside of the nucleus moving around in orbital "plauds". Scientists used to think that they orbited the nucleus in a ring like partners, but we now know electrons do not orbit. The electrons are so small and distant from the nucleus that must of the atom is just amply space!

Electrons have a regative charge. An atoms has an overall charge of zero when the number of negative electrons and positive protons are equal, trouverer, many chamical reactions involve changing the number of electrons of an atom. When electrons are gained or lost, the atom is called an ion. The number of neutrons can be different too between atoms of the same element, Asoms of the same element with different numbers of neutrons are called instopes.

The number of protons never changes. The amount of protons in the exclusiv, called the atomic number, is what gives an atom its identity as

a particular element. Each element on the Portotic Table has a different number of protons, and it is the basis for him the elements are arranged.



Employe Comprehension

- What are the three particles within an atom?
- 2, Which two particles are found in the nucleus?
- 2. Where are the electrons found in an aroun?
- a. Which particle gives an atom its identity as a perticular element?
- 5, move are elements arranged on the Periodic Table?

Ch 4 Guided Reading Key Atomic Structure

Judd E. Hollander

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