

CH7 CHEMICAL REACTIONS

- ★ Writing chemical equation
- ★ Balancing equations
- ★ Types of chemical Rxns

→ combination
→ Decomposition
→ Replacement
→ Combustion

STATES

(aq) = soluble in H_2O
= aqueous

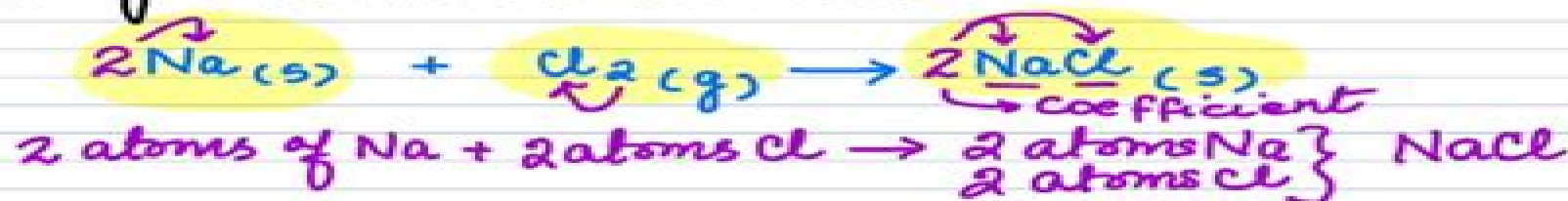
(s) = solid

(l) = liquid state

(g) = gas state

INTRODUCTION:

Writing a chemical Reaction -



CHEMICAL Reaction

- ★ permanent change = **STABLE** product
- ★ New substance, new properties set of physical + chem
- ★ gas → Bubbling, fizzing, fumes, gas is formed evolution of gas
- ★ solid → precipitate (ppt) formation precipitation
- ★ Heat → taken in exchange of energy
 - test tube feels cold
 - Given away released
 - test tube feels hot
- ★ color change - not always an indicator
- ★ Definite Ratios → Reactant Combine

explosion =
SOUND
LIGHT + HEAT
released

nuclear: gamma
+ radiation
released

Ch Redox Reactions Teacher Edition

Lei Shi



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