Balancing & Reactions Test SCH3U

Name:

___K __T __C __A Assessment of Learning (Product)
10 20 6 8

Multiple Choice: Circle the most correct answer. (10K)

State the type of reaction: 2 AlBr₂ + 3 Cl₂ → 2 AlCl₂ + 3 Br₂

A) synthesis B) decomposition C) single displacement

D) double displacement E) combustion F) no reaction

2. In what kind of reaction do two or more substances combine to form a new compound?

A) decomposition reaction C) double-displacement reaction

B) single displacement D) synthesis reaction

The reaction represented by the equation 2KClO₃(s) → 2KCl(s) + 3O₂(g) is a(n).

A) synthesis reaction. C) combustion reaction.

B) decomposition reaction.
 D) double-displacement reaction.

4. Which of the following terms appears when the eq. PbO₂ → PbO + O₂ is balanced?

A) PbO₅, B) 3 O₅, C) 3 PbO₇, D) O₇

Which of the terms appears when the eq. Cr₂S₃ + HCl → CrCl₃ + H₂S is balanced?

A) 3 HCl B) CrCl₂ C) 3 H₂S D) 2 Cr₂S₂

According to the Law of Conservation of Mass, in any chemical change

A) the mass of reactants is greater than the mass of products

B) the mass of reactants is less than the mass of products

C) the mass of reactants is equal to the mass of products

D) the masses of products and reactants have no predictable relationship

7. In a combustion reaction, one of the reactants is

A) hydrogen B) nitrogen C) oxygen D) a metal

8. When heated, a metal carbonate decomposes into a metal oxide and

A) carbon B) carbon dioxide C) oxygen D) hydrogen

9. Oxides of active metals, such as CaO, react with water to produce

A) metal carbonates. B) metal hydrides. C) acids. D) metal hydroxides.

10. When an insoluble solid compound is produced in a double-displacement reaction, a

A) gas bubbles off C) combustion reaction takes place

B) precipitate is formed D) halogen is produced

Ch11 Chemical Reactions Test A Answer Key

R Sandford

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