

## Chemical Reactions

### Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- |               |   |
|---------------|---|
| <u>yes</u>    | 1. A tomato smells rotten.  |
| <u>no</u>     | 2. A drinking glass breaks into smaller pieces.                     |
| <u>no</u>     | 3. A piece of ice melts.  |
| <u>yes</u>    | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u>    | 5. Candle wax burns.  |
| <u>yes/no</u> | 6. Molten candle wax solidifies.                                    |
| <u>yes</u>    | 7. Green leaves turn yellow and red as the seasons change.          |
| <u>yes</u>    | 8. Baking powder produces a gas that makes a cake rise.             |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

# Ch2study Guide Answer Key For Chemistry

**Arshad Iqbal**



## **Ch2study Guide Answer Key For Chemistry:**

The book delves into Ch2study Guide Answer Key For Chemistry. Ch2study Guide Answer Key For Chemistry is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Ch2study Guide Answer Key For Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Ch2study Guide Answer Key For Chemistry
- Chapter 2: Essential Elements of Ch2study Guide Answer Key For Chemistry
- Chapter 3: Ch2study Guide Answer Key For Chemistry in Everyday Life
- Chapter 4: Ch2study Guide Answer Key For Chemistry in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Ch2study Guide Answer Key For Chemistry. This chapter will explore what Ch2study Guide Answer Key For Chemistry is, why Ch2study Guide Answer Key For Chemistry is vital, and how to effectively learn about Ch2study Guide Answer Key For Chemistry.
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6. In chapter 5, the author will draw a conclusion about Ch2study Guide Answer Key For Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Ch2study Guide Answer Key For Chemistry.

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