

1. Write the equations depicting the behavior of the following acids and bases in water.

- $\text{HBr} \rightarrow \text{H}^+ + \text{Br}^-$
- $\text{HCN} \rightarrow \text{H}^+ + \text{CN}^-$
- $\text{RbOH} \rightarrow \text{Rb}^+ + \text{OH}^-$
- $\text{KOH} \rightarrow \text{K}^+ + \text{OH}^-$

2. Classify each reactant and product as an acid (A), base (B), conjugate acid (CA), or conjugate base (CB).

- $\text{F}^- + \text{H}_2\text{O} \rightarrow \text{HF} + \text{OH}^-$
- $\text{HClO} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{ClO}^-$
- $\text{H}_3\text{PO}_4 + \text{NH}_3 \rightarrow \text{NH}_4^+ + \text{H}_2\text{PO}_4^-$
- $\text{HNO}_3 + \text{HS}^- \rightarrow \text{H}_2\text{S} + \text{NO}_3^-$

a- base, acid, conjugate acid, conjugate base
b- acid, base, conjugate acid, conjugate base
c- acid, base, conjugate acid, conjugate base
d- acid, base, conjugate acid, conjugate base

3. Classify each of the following acids as monoprotic, diprotic, triprotic, or tetraprotic.

- H_2SO_4 (sulfuric acid) diprotic

- Acetic acid, show to the right



monoprotic

- HCN (cyanic acid) monoprotic

4. Oxaloacetic acid, which is produced in metabolic reactions in the human body has the following structure.



- Classify the above structure as a mono-, di-, tri-, or tetraprotic acid. Explain your answer.

Diprotic acid because it can only transfer 2 hydrogen protons

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Darrell D. Ebbing, David Bookin



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Table of Contents Ch 9 Practice Problems Chemistry Answers

1. Understanding the eBook Ch 9 Practice Problems Chemistry Answers
 - The Rise of Digital Reading Ch 9 Practice Problems Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ch 9 Practice Problems Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ch 9 Practice Problems Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ch 9 Practice Problems Chemistry Answers
 - Personalized Recommendations

- Ch 9 Practice Problems Chemistry Answers User Reviews and Ratings
- Ch 9 Practice Problems Chemistry Answers and Bestseller Lists
- 5. Accessing Ch 9 Practice Problems Chemistry Answers Free and Paid eBooks
 - Ch 9 Practice Problems Chemistry Answers Public Domain eBooks
 - Ch 9 Practice Problems Chemistry Answers eBook Subscription Services
 - Ch 9 Practice Problems Chemistry Answers Budget-Friendly Options
- 6. Navigating Ch 9 Practice Problems Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ch 9 Practice Problems Chemistry Answers Compatibility with Devices
 - Ch 9 Practice Problems Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ch 9 Practice Problems Chemistry Answers
 - Highlighting and Note-Taking Ch 9 Practice Problems Chemistry Answers
 - Interactive Elements Ch 9 Practice Problems Chemistry Answers
- 8. Staying Engaged with Ch 9 Practice Problems Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ch 9 Practice Problems Chemistry Answers
- 9. Balancing eBooks and Physical Books Ch 9 Practice Problems Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ch 9 Practice Problems Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ch 9 Practice Problems Chemistry Answers
 - Setting Reading Goals Ch 9 Practice Problems Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ch 9 Practice Problems Chemistry Answers
 - Fact-Checking eBook Content of Ch 9 Practice Problems Chemistry Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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