

CHAPTER 1

CHEMISTRY: THE STUDY OF CHANGE

Problem Categories

Biological: 1.24, 1.48, 1.69, 1.70, 1.78, 1.84, 1.93, 1.95, 1.96, 1.97, 1.105.

Conceptual: 1.3, 1.4, 1.11, 1.12, 1.15, 1.16, 1.54, 1.62, 1.89, 1.101, 1.103.

Environmental: 1.70, 1.87, 1.89, 1.92, 1.98.

Industrial: 1.51, 1.55, 1.72, 1.81, 1.91.

Difficulty Level

Easy: 1.3, 1.11, 1.13, 1.14, 1.15, 1.21, 1.22, 1.23, 1.24, 1.25, 1.26, 1.29, 1.30, 1.31, 1.32, 1.33, 1.34, 1.54, 1.55, 1.63, 1.64, 1.77, 1.80, 1.84, 1.89, 1.91.

Medium: 1.4, 1.12, 1.16, 1.35, 1.36, 1.37, 1.38, 1.39, 1.40, 1.41, 1.42, 1.43, 1.44, 1.45, 1.46, 1.47, 1.48, 1.49, 1.50, 1.51, 1.52, 1.53, 1.56, 1.57, 1.59, 1.60, 1.61, 1.62, 1.70, 1.71, 1.72, 1.73, 1.74, 1.75, 1.76, 1.78, 1.79, 1.81, 1.82, 1.83, 1.85, 1.94, 1.95, 1.96, 1.97, 1.98.

Difficult: 1.58, 1.65, 1.66, 1.67, 1.68, 1.69, 1.86, 1.87, 1.88, 1.90, 1.92, 1.93, 1.99, 1.100, 1.101, 1.102, 1.103, 1.104, 1.105, 1.106.

- 1.3**
- (a) Quantitative. This statement clearly involves a measurable distance.
 - (b) Qualitative. This is a value judgment. There is no numerical scale of measurement for artistic excellence.
 - (c) Qualitative. If the numerical values for the densities of ice and water were given, it would be a quantitative statement.
 - (d) Qualitative. Another value judgment.
 - (e) Qualitative. Even though numbers are involved, they are not the result of measurement.
- 1.4**
- (a) hypothesis
 - (b) law
 - (c) theory
- 1.11**
- (a) Chemical property. Oxygen gas is consumed in a combustion reaction; its composition and identity are changed.
 - (b) Chemical property. The fertilizer is consumed by the growing plants; it is turned into vegetable matter (different composition).
 - (c) Physical property. The measurement of the boiling point of water does not change its identity or composition.
 - (d) Physical property. The measurement of the densities of lead and aluminum does not change their composition.
 - (e) Chemical property. When uranium undergoes nuclear decay, the products are chemically different substances.
- 1.12**
- (a) Physical change. The helium isn't changed in any way by leaking out of the balloon.
 - (b) Chemical change in the battery.
 - (c) Physical change. The orange juice concentrate can be regenerated by evaporation of the water.
 - (d) Chemical change. Photosynthesis changes water, carbon dioxide, etc., into complex organic matter.
 - (e) Physical change. The salt can be recovered unchanged by evaporation.

Chang Chemistry Problem Solutions

**Philip C. Keller, Raymond Chang, Jill L.
Keller**



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