



Chapter 12 Stoichiometry Notes Packet

Big Picture Ideas

1. Stoichiometry is the quantitative relationship between the masses of reactants and products in a chemical reaction.
2. Stoichiometry is a quantitative relationship between the masses of reactants and products in a chemical reaction.

Big Picture Questions

1. How can stoichiometry be used to determine the mass of a product from the mass of a reactant?
2. How can stoichiometry be used to determine the mass of a reactant from the mass of a product?

Suggested Resources

Stoichiometry: A Quantitative Approach
Laboratory Manual

Stoichiometry: A Quantitative Approach
Laboratory Manual
Stoichiometry: A Quantitative Approach

Key Terms

1. Stoichiometry	2. Stoichiometric Coefficient	3. Stoichiometric Ratio
4. Stoichiometric Coefficient	5. Stoichiometric Ratio	6. Stoichiometric Ratio
7. Stoichiometric Ratio	8. Stoichiometric Ratio	9. Stoichiometric Ratio
10. Stoichiometric Ratio	11. Stoichiometric Ratio	12. Stoichiometric Ratio

Stoichiometry is the quantitative relationship between the masses of reactants and products in a chemical reaction.

Stoichiometric Coefficient is the number of molecules of a substance that are involved in a chemical reaction.

1. Stoichiometry is the quantitative relationship between the masses of reactants and products in a chemical reaction.
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3. Stoichiometric Ratio is the ratio of the masses of reactants and products in a chemical reaction.

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8. Stoichiometric Ratio is the ratio of the masses of reactants and products in a chemical reaction.

Experiment	1. Stoichiometry	2. Stoichiometry	3. Stoichiometry	4. Stoichiometry
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