

1. Draw the structures of the following compounds

- cyclopentadiene (explain why the numbers for the double bonds (1,3-), which are formally required, are usually omitted; i.e., why is only one isomer possible; why not 1,2- or 1,4-cyclopentadiene?)
- 1,3,5-cycloheptatriene
- (2E, 4Z)-2,4-heptadiene
- isoprene (look it up and also give its IUPAC name)

2. Draw a molecule that has the following characteristics.

- a decalin (2 six-member rings joined side by side) with both conjugated and non-conjugated double bond (three double bond total)
- a compound containing a 5-member ring and conjugated double bonds that are locked in an s-trans orientation
- a eight-member ring containing two cumulated double bonds (i.e., an allene)
- a decalin with two double bonds that are locked in an s-cis orientation

3.

- Draw isoprene in both the s-cis and the s-trans conformation
- Draw all trans and all cis 2,4,6-octatriene
- Draw all trans 1,3,5,7-octatetraene in all possible conformations (rotamers, not geometric isomers) where the four double bonds are all conjugated (planar)

4.

- Fill in the missing diene that would lead to this isomer as the thermodynamic, but not the kinetic, product.



- Show the mechanism for the above reaction, including the formation of both kinetic (low temperature) and thermodynamic (higher temperature) products.
- Show the mechanism for the conversion of the kinetic to the thermodynamic product.
- Show a reasonable energy curve for this transformation and draw in all structures, including intermediates and transition states

5. Draw an example of the following.

- an acyclic, conjugated diene that cannot be s-cis due to substitution at C2 and C3 of the diene
- an acyclic, conjugated diene that cannot be s-cis due to substitution at C1 and C4 of the diene

6. Rank the following alkenes from least reactive with 1,3-butadiene to most reactive. Explain the ranking of each compound with a couple of words.



# Chap 16 Review Answers Chem

**Ralph Thomas Myers**



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