

**11.1 - Describing Chemical Reactions**

1. What two substances reacted in the Hindenburg crash?
2. In general, what happens in a **chemical reaction**?
3. Look at Figure 11.1. What are some of the evidences of **chemical change** (a.k.a. chemical reaction)?
4. Give an example of a **word equation** for a chemical reaction (you can make one up or choose one from the book).
5. How is a **chemical equation** different from a word equation?
6. What is missing in a skeleton equation?
7. What symbols are used to represent the phases of substances in chemical reactions?
8. What does the symbol (aq) mean? \_\_\_\_\_
9. Use the space below to do Practice Problems #1 and #2.  
1) \_\_\_\_\_ 2) \_\_\_\_\_
10. A complete description of a reaction must include \_\_\_\_\_,  
but also \_\_\_\_\_.
11. Define the following:
  - a. **coefficients** -
  - b. **subscripts** -
  - c. **balanced equation** -
12. How does the **law of conservation of mass** relate to balancing reaction equations?
13. In a chemical reaction, what **MUST** be the same for the reactants and the products?  
(There are two great answers to this question!)

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## Describing Chemical

**E Durkheim**



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