

11 Chemical Reactions

Big idea REACTIONS

11.1 Describing Chemical Reactions

For students using the Foundation edition, assign problems 1–17.

Essential Understanding Chemical reactions are represented by balanced chemical equations.

Lesson Summary

Introduction to Chemical Equations A chemical equation uses symbols, and sometimes words, to show the reactants and products of a chemical reaction.

- ▶ A skeleton equation uses chemical formulas to represent reactants and products, but it does not indicate the relative amounts of each.
- ▶ Anything that enters into a reaction, such as heat or a catalyst, but is not a reactant or product is shown above or below the yields arrow in the equation.

Balancing Chemical Equations Chemical equations are balanced to show that mass is conserved during chemical reactions.

- ▶ A balanced equation shows the relative amounts of reactants and products, and it contains equal numbers of each type of atom on both sides of the equation.
- ▶ Chemical equations are balanced by using coefficients in front of the chemical formulas for the reactants and the products in a skeleton equation.



BUILD Math Skills

Balancing Equations All chemical equations must be balanced because of the law of conservation of mass, which states that matter cannot be created or destroyed. So, the number of atoms that you start with at the beginning of the reaction must equal the number of atoms that you end up with.

For example, the reaction $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ follows the law of conservation of mass because you start with 2 magnesium atoms and 2 oxygen atoms and you end up with 2 magnesium atoms and 2 oxygen atoms. You can think of the \rightarrow as an $=$ sign.



Turn the page to learn more about balancing equations.

Chapter 11 Describing Chemical Reactions Answer Key

Daniel F McAuley



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LSENS, a General Chemical Kinetics and Sensitivity Analysis Code for Homogeneous Gas-phase Reactions. 2: Code Description and Usage Krishnan Radhakrishnan, 1994 **Physical and Chemical Hydrogeology** Patrick A. Domenico, Franklin W. Schwartz, 1997-10-15 This book shows readers how to apply hydrogeology principles to a host of problems related to water supply contamination and energy resources It discusses hydraulic testing modeling of contaminant transport process and parameter determination and remediation It also addresses porosity permeability and flow for continental environments marine environments and the borders between them *Chemistry* James N. Spencer, George M. Bodner, Lyman H. Rickard, 2002-02-28 The second edition of Spencer's Chemistry Structure and Dynamics has been the most successful reform project published for the General Chemistry course The authors have revised the text by building on the recommendations of the ACS's Task Force on the General Chemistry Curriculum and suggestions from the adopters of the first edition This innovative text provides a fifteen chapter introduction to the fundamental concepts of Chemistry A collection of additional topics at the end of each chapter allow instructors to supplement and tailor their courses according to individual need Three major themes link the content of the book the process of science the relationship between molecular structure and physical chemical properties and the relationship between the microscopic and macroscopic levels **Chaos Theory** Kais A. M. Al Naimee, 2018-03-28 With a good background in nonlinear dynamics chaos theory and applications the author of this leading book gives a systematic treatment of the basic principle of nonlinear dynamics in different fields The contributions from leading international scientists active in the field provide a comprehensive overview of our current level of background on chaos theory and applications in different sciences In addition they show overlap with the traditional field of control theory in scientific community **Prentice Hall Science Explorer** Michael J. Padilla, 2002 Experimental science. Physiography, section 1, by R.A. Gregory and A.T. Simmons sir Richard Arman Gregory (bart.), 1899 **Focus on Physical Science California Edition** Michael J. Padilla, 2007 *Fundamentals of Groundwater* Franklin W. Schwartz, Hubao Zhang, 2024-01-18 Fundamentals of Groundwater A thoroughly updated classic on the fundamentals of groundwater The second edition of Fundamentals of Groundwater delivers an expert discussion of the fundamentals of groundwater in the hydrologic cycle and applications to contemporary problems in hydrogeology The theme of the book is groundwater broadly defined and it covers the theory and practice of groundwater from basic principles of physical and chemical hydrogeology to their application in traditional and emerging areas of practice This new edition contains extensive revisions including new discussions of human impacts on aquifers and strategies and concepts for sustainable development of groundwater It also covers the theory of groundwater flow including concepts of hydraulic head and the Darcy equation and ground water surface water interactions as well as geochemistry and contamination Readers will also find A thorough introduction to the techniques of water resource investigations and regional groundwater flow Comprehensive explorations

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









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