

## Chapter

# 10

# ELECTROCHEMISTRY

## MULTIPLE CHOICE QUESTIONS

- Reduction always takes place: (LHR 05, 12)
  - At anode
  - At cathode
  - At both electrodes
  - does not occur at electrodes
- The electrolyte KOH is used in cell: (GRW 06)
  - Lead accumulator
  - Ni – Cd cell
  - Alkaline battery
  - Silver oxide battery
- In the reaction  $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$ : (BWP 10, GRW 09, LHR 06)
  - Fe is reduced
  - Fe is oxidized
  - $\text{Cl}_2$  is oxidized
  - None of these happens
- Oxidation state of hydrogen in  $\text{CaH}_2$  is: (GRW 07)
  - +1
  - 1
  - +2
  - zero
- Cu metal can be purified in electrolytic cell by making the impure Cu as: (GRW 07)
  - Anode
  - Cathode
  - Anode and Cathode
  - Depends upon nature of solution
- Loss of electrons is called: (LHR 07)
  - Oxidation
  - Reduction
  - Hydration
  - Dehydration
- Fuel cells convert chemical energy into: (GRW 07)
  - Heat energy
  - Light energy
  - Electrical energy
  - Mechanical energy
- Electrolysis is used for: (LHR 08)
  - Electroplating
  - Manufacture of sodium metal
  - Manufacture of Al
  - All of these
- Nelson's cell and Down's cell are example of: (FSD 09, GRW 08)
  - Electrochemical cell
  - Galvanic cell
  - Electrolytic cell
  - None of these
- The oxidation number of Cr in  $\text{K}_2\text{Cr}_2\text{O}_7$  is: (GRW 09, 10, LHR 10, 12, 13)
  - +14
  - +12
  - +6
  - +13
- The electrolyte used in fuel cell is: (LHR 09, DGK, GRW 10)
  - Aqueous  $\text{NaCl}$
  - Molten  $\text{NaCl}$
  - $\text{KOH}$
  - $\text{H}_2\text{SO}_4$

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In a global driven by information and connectivity, the power of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Chapter 10 Electrochemistry Mcqs Out Class**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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