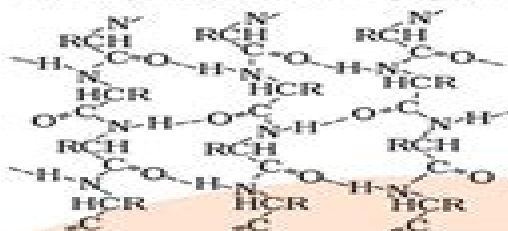


# Chemistry

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(Chapter 10) (Biomolecules) (Exercises)  
(Class 12)

maximum extension and then laid side by side. These peptide chains are held together by intermolecular hydrogen bonds.



**Question 10.14:**

What type of bonding helps in stabilising the  $\alpha$ -helix structure of proteins?

**Answer 10.14:**

The H-bonds formed between the  $-\text{NH}$  group of each amino acid residue and the  $>\text{C}=\text{O}$  group of the adjacent turns of the-helix help in stabilising the helix.

**Question 10.15:**

Differentiate between globular and fibrous proteins.

**Answer 10.15:**

Fibrous protein	Globular protein
1. It is a fibre-like structure formed by the polypeptide chain. These proteins are held together by strong hydrogen and disulphide bonds.	The polypeptide chain in this protein is folded around itself, giving rise to a spherical structure.
2. It is usually insoluble in water.	It is usually soluble in water.
3. Fibrous proteins are usually used for structural purposes. For example, keratin is present in nails and hair; collagen tendons; and myosin in muscles.	All enzymes are globular proteins. Some hormones such as insulin are also globular proteins.

**Question 10.16:**

How do you explain the amphoteric behaviour of amino acids?

**Answer 10.16:**

In aqueous solution, the carboxyl group of an amino acid can lose a proton and the amino group can accept a proton to give a dipolar ion known as zwitter ion.



Therefore, in zwitter ionic form, the amino acid can act both as an acid and as a base.



Thus, amino acids show amphoteric behaviour.

# Chapter 10 Solutions Manual Chemistry

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