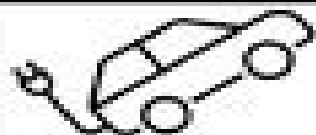




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Write the following vocabulary words and definitions from above your to

1. Avogadro's Number - The number of representative particles in a mole. 6.02×10^{23}
2. Mole - The SI base unit used to measure the amount of substance. It is the amount of a pure substance that contains as many particles as there are atoms in 12 grams of carbon-12.
3. Empirical Formula - A formula that shows the simplest whole number ratio of atoms in a compound.
4. Molecular Formula - A formula that shows the actual number of atoms of each element in a molecule.
5. Percent Composition - The percentage by mass of each element in a compound.
6. Molar Mass - The mass of one mole of a substance, expressed in grams per mole (g/mol).
7. Molar Mass - The mass of one mole of a substance, expressed in grams per mole (g/mol).
8. Molar Mass - The mass of one mole of a substance, expressed in grams per mole (g/mol).
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11. Molar Mass - The mass of one mole of a substance, expressed in grams per mole (g/mol).
12. Molar Mass - The mass of one mole of a substance, expressed in grams per mole (g/mol).



A formula that specifies the actual number of atoms of each element.

- ☐ molar mass
- ☐ empirical formula
- ☐ molecular formula
- ☐ Avogadro's number

The percent by mass of each element in a compound.

- ☐ molar mass
- ☐ mass number
- ☐ molar mass
- ☐ percent composition

A chemical combination of elements that forms a new substance by chemical means.

- ☐ hydrate
- ☐ empirical formula
- ☐ compound



The Mole Vocabulary Quiz

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Number of representative particles in a mole

- ☐ molar mass
- ☐ empirical formula
- ☐ molecular formula
- ☐ Avogadro's number

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Chapter 11 Assessment The Mole Reviewing Vocabulary

RD Boyd



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 - Chapter 5: Conclusion
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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 11 Assessment The Mole Reviewing Vocabulary.

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