

Name _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____.
 - A) the solvent is a gas and the solute is a solid
 - B) they are accompanied by an increase in order
 - C) they are accompanied by an increase in disorder
 - D) they are accompanied by another process that is exothermic
 - E) the solvent is water and the solute is a gas
- 2) The phrase "like dissolves like" refers to the fact that _____.
 - A) polar solvents dissolve nonpolar solutes and vice versa
 - B) solvents can only dissolve solutes of similar molar mass
 - C) gases can only dissolve other gases
 - D) polar solvents dissolve polar solutes and nonpolar solvents dissolve nonpolar solutes
 - E) condensed phases can only dissolve other condensed phases
- 3) An unsaturated solution is one that _____.
 - A) has no double bonds
 - B) has a concentration lower than the solubility
 - C) contains no solute
 - D) contains the maximum concentration of solute possible, and is in equilibrium with undissolved solute
 - E) contains more dissolved solute than the solubility allows
- 4) A solution with a concentration higher than the solubility is _____.
 - A) is not possible
 - B) is saturated
 - C) is supercritical
 - D) is unsaturated
 - E) is supersaturated
- 5) Which one of the following substances would be the most soluble in CCl_4 ?
A) H_2O B) NaCl C) $\text{CH}_3\text{CH}_2\text{OH}$ D) NH_3 E) $\text{C}_{10}\text{H}_{22}$
- 6) Which one of the following is most soluble in water?
 - A) CH_3OH
 - B) $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$
 - C) $\text{CH}_3\text{CH}_2\text{OH}$
 - D) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$
 - E) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

Chapter 11 Chemistry Test

Stella Goostray, Walter Gerald Karr



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