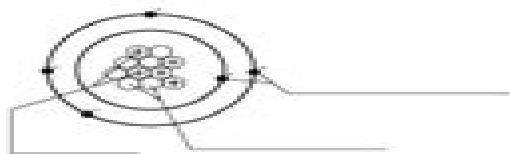


Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- What type of charge does a proton have? _____
- What type of charge does a neutron have? _____
- What type of charge does an electron have? _____
- Which two subatomic particles are located in the nucleus of an atom? _____

2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.

8	← Atomic Number
O	← Symbol
Oxygen	← Name
15.999	← Average atomic mass

6	#P _____
C	#N _____
Carbon	#E _____
12.011	

10	#P _____
Ne	#N _____
Neon	#E _____
20.108	

19	#P _____
K	#N _____
Potassium	#E _____
39.098	

3. Complete the table using your knowledge of the periodic table.

Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na			11		12
K		39		19	
			38		50
F				9	10
	20	40		20	
	50			50	69
I	53	127			

Chapter Test Atom And Periodic Table

Anthony S. Fauci



Chapter Test Atom And Periodic Table:

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Table of Contents Chapter Test Atom And Periodic Table

1. Understanding the eBook Chapter Test Atom And Periodic Table
 - The Rise of Digital Reading Chapter Test Atom And Periodic Table
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Test Atom And Periodic Table
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter Test Atom And Periodic Table
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Test Atom And Periodic Table
 - Personalized Recommendations
 - Chapter Test Atom And Periodic Table User Reviews and Ratings

- Chapter Test Atom And Periodic Table and Bestseller Lists
- 5. Accessing Chapter Test Atom And Periodic Table Free and Paid eBooks
 - Chapter Test Atom And Periodic Table Public Domain eBooks
 - Chapter Test Atom And Periodic Table eBook Subscription Services
 - Chapter Test Atom And Periodic Table Budget-Friendly Options
- 6. Navigating Chapter Test Atom And Periodic Table eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter Test Atom And Periodic Table Compatibility with Devices
 - Chapter Test Atom And Periodic Table Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter Test Atom And Periodic Table
 - Highlighting and Note-Taking Chapter Test Atom And Periodic Table
 - Interactive Elements Chapter Test Atom And Periodic Table
- 8. Staying Engaged with Chapter Test Atom And Periodic Table
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter Test Atom And Periodic Table
- 9. Balancing eBooks and Physical Books Chapter Test Atom And Periodic Table
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter Test Atom And Periodic Table
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter Test Atom And Periodic Table
 - Setting Reading Goals Chapter Test Atom And Periodic Table
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter Test Atom And Periodic Table
 - Fact-Checking eBook Content of Chapter Test Atom And Periodic Table
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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