

CHAPTER 11 REVIEW*Molecular Composition of Gases***MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
(a) polarity of the molecule
(b) pressure of the gas
(c) temperature of the gas
(d) number of moles in the sample
2. _____ Which of the following statements best represents Graham's law for two gases at the same temperature?
(a) The molar masses and the velocities are inversely proportional.
(b) The molar masses and the velocities are directly proportional.
(c) The molar masses and the square roots of the velocities are directly proportional.
(d) The molar masses and the squares of the velocities are inversely proportional.
3. Water is synthesized from its elements according to the following equation:
$$2\text{H}_2(g) + \text{O}_2(g) \rightarrow 2\text{H}_2\text{O}(l)$$

_____ a. Which is greater, the volume of H_2 consumed or the volume of O_2 consumed?
_____ b. Which is greater, the mass of H_2 consumed or the mass of O_2 consumed?
4. Consider a 2.24 L sample of an ideal gas at STP.
_____ a. How many moles are present in the sample, or is more data needed?
_____ b. How many molecules are present in the sample, or is more data needed?
_____ c. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

5. _____ Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

Chapter11 Review Gases Mixed Review

David A. Anderson



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Physics Holt Rinehart & Winston, 2000-12 **Applied Mechanics Reviews** , 1971 *Index of Specifications and Standards* , 2005 *Department Of Defense Index of Specifications and Standards Alphabetical Listing Part I* July 2005 , **Iron & Coal Trades Review** , 1927 Vol 115 includes Diamond jubilee issue 1867 1927 **Environmental Economics and Natural Resource Management** David A. Anderson, 2013-09-23 The tools of environmental economics guide policymakers as they weigh development against nature present against future and certain benefits against uncertain consequences From reluctant but necessary calculations of the value of life to quandaries over profits at the environment s expense the policies and research findings explained in this textbook are relevant to decisions made daily by individuals firms and governments The fourth edition of *Environmental Economics and Natural Resource Management* pairs the user friendly approaches of the previous editions with the latest developments in the field A story based narrative delivers clear concise coverage of contemporary policy initiatives To promote environmental and economic literacy we have added even more visual aids including color photographs and diagrams unmatched in other texts Ancillaries include an Instructor s Guide with answers to all of the practice problems and downloadable slides of figures and tables from the book The economy is a subset of the environment from which resources are obtained workers and consumers receive sustenance and life begins Energy prices and environmental calamities constrain economic growth and the quality of life The same can be said about overly restrictive environmental policies It is with an appreciation for the weighty influence of this discipline and the importance of conveying it to students that this textbook is crafted Bulletin of Thermodynamics and Thermochemistry , 1971 **The Chemical News and Journal of Physical Science** , 1908 *The Mining World Index of Current Literature* , 1912 *The Chemical News and Journal of Industrial Science* William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1878

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