

Name _____

Period _____

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Elements, Compounds, and Mixtures Vocabulary Practice

Directions: Use the word bank below to fill in the blanks for each definition. Answer all questions that follow using the vocabulary words and their definitions.

Matter	1. _____ - made from the same materials throughout and have the same chemical properties throughout.
Pure substance	2. _____ - not pure substances; they are physically combined, not chemically.
Physical properties	3. _____ - an element in its pure form that's always bonded to itself.
Chemical properties	4. _____ - this must be taken in or given off when a compound is broken apart or put together.
Elements	5. _____ - anything that has mass and takes up space.
Diatonic	6. _____ - particles settle and form layers over time.
Compounds	7. _____ - the parts of the mixture are, and will stay, evenly mixed.
Chemical formulas	8. _____ - characteristics such as reactivity, flammability, and acidity.
Mixtures	9. _____ - particles can be seen in a beam of light.
Homogenous	10. _____ - pure substances made of two or more elements that are chemically bonded together. It has different properties than the individual elements in it.
Heterogenous	11. _____ - represent how many atoms of each element are part of the compound.
Suspension	12. _____ - characteristics such as density, melting point, malleability.

Chapter Review Elements And Their Properties

Vocabulary Review

Clifford Smyth



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