#### Nelson Biology VCE Units 3 & 4

#### Answers to Chapter Questions

#### Chapter I

#### Review questions

- Both non-living and living things are composed of matter. Matter is made up of atoms.
- 2 Living things are capable of reproduction, growth and movement.
- 3 6 electrons
- 4 The proton defines the element.
- 5 The atomic number would be 15 and the mass number would be 32. The symbol for phosphorus is P. The standard notation is <sup>32</sup><sub>15</sub> P.
- 6 The number of neutrons may vary in all the isotopes of a particular element. Hence, an element may have different mass numbers.
- 7 The four major elements found in living things are oxygen, carbon, hydrogen and nitrogen.
- 8 Radioactive isotopes are used as a source of radiation in therapy and as tracers to tag particular compounds.
- 9 A compound is a pure substance made up of more that one type of atom held together by chemical bonds.
- 10 Ionic compounds are composed of cations and anions, which form ionic bonds. Molecular compounds result from atoms sharing their electrons to form covalent bonds. Molecular compounds vary in their solubility in water depending upon their polar nature. Ionic compounds are generally more soluble in water than molecular compounds.
- 11 Molecular compounds result from the sharing of electrons between non-metal elements, whereas ionic compounds are composed of ions and result from the transfer of electrons between metallic and non-metallic elements. As a result of the charges carried, ionic compounds tend to be more readily dissolved in water compared to molecular compounds.
- 12 A chemical bond is the force of attraction holding atoms together in chemical molecules.
- 13 Molecules that have permanent regions of opposite charge are termed 'polar'. Non-polar molecules have no permanent regions of opposite charge.
- 14 Water molecules are highly polar and attract one another, forming hydrogen bonds. Hydrogen bonding determines many of the unique properties of water.
- Hydrophilic molecules, such as alcohol, have regions of opposite charge that readily form hydrogen bonds with water molecules and hence dissolve.

# **Chapter Review Nelson Biology**

**Geological Survey (U.S.)** 

## **Chapter Review Nelson Biology:**

Cladistics David M. Williams, Malte C. Ebach, 2020-08-06 This new edition of a foundational text presents a contemporary review of cladistics as applied to biological classification. It provides a comprehensive account of the past fifty years of discussion on the relationship between classification phylogeny and evolution. It covers cladistics in the era of molecular data detailing new advances and ideas that have emerged over the last twenty five years Written in an accessible style by internationally renowned authors in the field readers are straightforwardly guided through fundamental principles and terminology Simple worked examples and easy to understand diagrams also help readers navigate complex problems that have perplexed scientists for centuries This practical guide is an essential addition for advanced undergraduates postgraduates and researchers in taxonomy systematics comparative biology evolutionary biology and molecular biology

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erweiterte Kapitel u a zu folgenden Themen An sthesie und Komplikationen bei Eingriffen an der Wirbels ule auch postoperative Erblindung Robotergest tzte Operationsverfahren bei Eingriffen am kindlichen Urinaltrakt An sthesie bei Eingriffen an Personen mit angeborenen Herzerkrankungen Dieses neue Kapitel behandelt keine Herzoperationen Umfangreiche neue Ultraschallbilder bei rtlicher Bet ubung Wiederbelebung bei Neugeborenen Betreuung und Rekonvaleszenz von Kindern nach chirurgischen Eingriffen neues Kapitel Auch die 6 Auflage von Gregory s Pediatric Anesthesia ist ein verl sslicher und zug nglicher Leitfaden f r An sthesisten die j ngere Patienten betreuen

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