

## CHAPTER 7: GASES, LIQUIDS, AND SOLIDS

CHE 120

### CHAPTER 07 GASES, LIQUIDS and SOLIDS Practice Questions

- Which of the following statements is not a part of kinetic molecular theory?
  - Matter is composed of particles that are in constant motion.
  - Particle velocity increases as the temperature increases.
  - Particles in a system cannot transfer energy to each other.
  - Particle interactions involve electrostatic attractions and repulsions.
- In which of the following groupings are the three terms closely related?
  - kinetic energy, energy of motion, cohesive forces
  - potential energy, energy of attraction, disruptive forces
  - kinetic energy, electrostatic interactions, disruptive forces
  - potential energy, electrostatic interactions, cohesive forces
- In the liquid state, disruptive forces are
  - roughly of the same magnitude as cohesive forces
  - very weak compared to cohesive forces
  - dominant over cohesive forces
  - unimportant
- The phrases "particles close together and held in fixed positions" and "completely fills the container" apply, respectively, to
  - liquids and solids
  - solids and gases
  - gases and liquids
  - liquids and gases
- How many times larger in size is the atmosphere pressure unit (atm) than the mm Hg pressure unit?
  - 100
  - 325
  - 760
  - 1000
- Boyle's law involves which of the following?
  - a direct proportion
  - a constant volume
  - a constant pressure
  - a constant temperature
- Indicate what the missing words are in the following statement of Charles's law: At constant pressure, the volume of a gas sample is \_\_\_\_\_ proportional to its \_\_\_\_\_ temperature.
  - directly; Celsius
  - directly; Kelvin
  - inversely; Celsius
  - inversely; Kelvin

# Chapter Solids Liquids And Gases Test

**AN Whitehead**



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